

Synthesis of fluorinated chiral amines using *N-tert*-butylsulfinyl imines

Fluorinated chiral amines are highly important building blocks in medicinal chemistry since fluorine decreases the basicity of the amine functionality and, thus, improves the bioavailability of a drug molecule. Recently, stereoselective synthesis of fluorinated chiral amines based on the use of *N-tert*-butylsulfinyl imines has attracted much attention due to the high stereoselectivity, broad substrate scope and easy experimental handing. In this article, we summarize developments in the field over the last decade, including the synthesis of trifluoromethylated, difluoromethylated, *gem*-difluoromethylenated and monofluoromethylated chiral amines. Stereoselective synthesis of fluorinated chiral amines based on the use of *N-tert*-butylsulfinyl imines has been accomplished by two major strategies. One is the stereoselective addition or asymmetric reduction of fluorinated *N-tert*-butylsulfinyl imines, the development of which is lacking new efficient and atom-economic fluorinated reagents.

Fluorinated amines, especially chiral fluorinated amines, are important synthetic building blocks in drug design because of the minimal structural changes but maximal shifts in electron distribution caused by fluorine [1,2]. Furthermore, the basicity of amines (especially the β-fluorinated amines) can be tuned by fluorine or fluoroalkyl groups, thus increasing high bioavailability of a target drug [3-5]. Over the past three decades, efficient stereoselective synthesis of fluorinated amines has attracted much attention; such synthetic approaches are based mainly on the use of fluorinating reagents (e.g., diethylaminosulfur trifluoride and the hydrogen fluoride-pyridine complex) and fluorinated building blocks (e.g., fluorinated imidoyl halides and fluorinated α -imino esters and imines) [6-8]. In this context, diastereoselective synthesis of fluorinated amines using N-tert-butylsulfinyl imines has become one of the most important and widely used approaches. Although enantiopure N-tertbutylsulfinyl imines were firstly reported by García Ruano and co-workers in 1996 [9], their wide application in asymmetric synthesis was initiated by Ellman's seminal work on the practical preparation of N-tert-butylsulfinyl imines by condensation of optically pure N-tertbutylsulfinamide with aldehydes or ketones [10-18]. Furthermore, the easy recycling of the tert-butylsulfinyl chiral auxiliary group (by conversion back to enantiopure tert-butylsulfinamide) greatly improves the practicality and large-scale applications of tert-butylsulfinamidebased amine synthesis chemistry [19]. The fact

that the *tert*-butylsulfinyl group activates the imine functional group (for the addition of many different classes of nucleophiles) and serves as a powerful chiral-directing group makes it highly attractive in the asymmetric synthesis of fluorinated chiral amines. In this article, we would like to summarize the stereoselective synthesis of fluorinated chiral amines based on the use of Ellman's *N-tert*-butylsulfinyl imines developed during the last decade, including the synthesis of trifluoromethylated, difluoromethylated, *gem*difluoromethylenated and monofluoromethylated chiral amines.

Fluorinated chiral amines from fluorinated tert-butylsulfinyl imines & their derivatives

One of the useful methods for asymmetric synthesis of fluorinated chiral amines is based on the use of fluorinated enantiopure *tert*-butylsulfinyl imines and their derivatives as precursors [20]. It should be noted that the isolation and characterization of fluorinated enantiopure *tert*butylsulfinyl imines are generally difficult due to their susceptibility to hydrolysis [21]. Therefore, in most cases, fluorinated *tert*-butylsulfinyl imines are generated *in situ* to react with various electrophiles [22,23].

In 2006, Lu and co-workers described the solvent-controlled asymmetric Strecker reactions with unstable trifluoromethyl sulfinimines 1, and each diastereomer (2 or 3) of the α -trifluoromethyl α -cyano amines was obtained in good yield and with high diastereoselectivity

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FLUORINATED AMINE

Amine compounds containing fluorine atom(s), with the fluorine substitution often on the β -position to the amino functionality

STEREOSELECTIVE SYNTHESIS

A chemical reaction (or reaction sequence) that produces the stereoisomeric (enantiomeric, diastereomeric or Z-/E-isomeric) products in unequal quantities

N-TERT-BUTYLSULFINYL IMINE

A class of imine compounds that possess a (generally homochiral) *tert*-butylsulfinyl group on the nitrogen atom of imino group



and IMSCN.							
$\begin{array}{c} & \bigcirc & R & \bigcirc & F_3 & \underbrace{TMSCN, -35^\circ C}_{DMF} & \bigcirc & \bigcap & R & \underbrace{TMSCN, rt}_{S^\circ} & \bigwedge & \bigcap & CF_3 & \underbrace{TMSCN, rt}_{Hexane} & \bigcirc & R & \bigcirc & CF_3 \\ & & & & & & & \\ & & & & & & & \\ & & & & & & & \\ & & & & & & & \\ & & & & & & & \\ & & & & & & & \\ & & & & & & & \\ & & & & & & & \\ & & & & & & & \\ & & & & & & & \\ & & & & & & \\ & & & & & & \\ & & & & & & \\ & & & & & & \\ & & & & & & \\ & & & & & & \\ & & & & & & \\ & & & & & & \\ & & & & & & \\ & & & & & & \\ & & & & & & \\ & & & & & & \\ & & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & \\ & & & & & \\ & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & & \\ & & & & & & \\ & & & & & & \\ & & & & & & \\ & & & & & & \\ & & & & & & \\ & & & & & & \\ & & & & & & \\ & & & & & & \\ & & & & & & \\ & & & & & & \\ & & & & & & \\ & & & & & & \\ & & & & & & \\ & & & & & & \\ & & & & & & \\ & & & & & & \\ & & & & & & \\ & & & & &$							
Entry	Substrate	Hexane	e	DMF			
		Total yield (%)	dr (2/3)	Total yield (%)	dr (2/3)		
1	R = Me	69	27:1	71	1:19		
2	R = Et	77	7:1	76	1:10		
3	$R = n - C_6 H_{13}$	88	14:1	84	1:11		
4	$R = CH_2Bn$	92	7:1	89	1:15		
5	R = Ph	85	99:1	72	1:6		
6	$R = p - MeC_6H_4$	87	11:1	69	1:7		
7	$R = p - MeOC_6H_4$	89	14:1	78	1:9		
8	$R = p - CIC_6H_4$	83	8:1	71	1:6		
TMSCN: Data from	Trimethylsilyl cyanide. n [21].						

Table 1. Asymmetric Strecker reaction between CF₂-sulfinimines (1)



Figure 1. Different transition state models.

Table 2. Asymmetric addition reactions of Grighard reagents.					
F ₃	Q S NH R-M C OEt 4		0 Š N CF ₃	R-M	Q S NH F ₃ C R
Entry	RM	Solvent	Product	dr [(S _s ,S):(S _s ,R)]	Yield (%)
1	MgBr	CH_2CI_2	6 A	22:1	77
2	MgBr	CH ₂ Cl ₂	6B	11:1	81
3	MgBr	CH_2CI_2	6C	5:1	98
4	MgBr	$\rm CH_2\rm Cl_2$	6D	>30:1	74
5	MgBr	CH ₂ Cl ₂	6E	6:1	98
6	MgBr	THF	6E	3.5:1	97
Data from	m [22].				

(TABLE 1) [21]. The authors rationalized the stereoselectivity using different models. The Strecker reaction in hexane may proceed via the six-membered chair-like model and the sulfinyl groupactivated trimethyl cyanide (TMSCN) giving the (*S*, *Rs*)-isomer **2** as the predominant product. However, the Strecker reaction in DMF may undertake Fujisawa's model, in which DMF serves not only as a polar solvent but also as a Lewis base to active TMSCN. Therefore, (*R*, *Rs*)-isomer **3** was obtained as the major product (FIGURE 1). Further deprotection and hydrolysis afforded α -trifluoromethyl α -amino acid [21].

Later, the addition of Grignard reagents to trifluoromethyl *tert*-butylsulfinimine—ethanol adducts was reported by Kuduk and co-workers (TABLE 2) [22]. In these reactions, sulfinamino acetals 4 served as stable precursors for the *in situ* generation of imines 5, which underwent reactions with Grignard reagents to afford allyltrifluoromethylamines 6 in good yields and with moderate-to-excellent diastereoselectivity.

Truong and co-workers reported the synthesis of a variety of optically enriched 1-aryl-2,2,2-trifluoroethylamine analogs 7 in moderate-to-good yields using 2-methyl-N-(2,2,2-trifluoro-ethylidene)propane-2-sulfinamide 5 as a key precursor (TABLE 3) [23]. The addition of phenyllithium to imine 5 appeared to have proceeded via a nonchelated transition-state model where phenyllithiun preferably added to the imine from the less-hindered face (derived from the most stable conformation of the imine) to afford the major diastereomer adduct (Ss,S)-7a [23]. It should be pointed out that recent studies have shown that the more stable conformation of the reacting N-tert-butylsulfinimines is the conformation where the N-S-O bonds are coplanar and the S-O bond lies away from the lone pair of the nitrogen atom [16,24-26].

Imine **5** was also subjected to the rhodiumcatalyzed 1,2-addition of aryboronic acids affording optically enriched sulfinamides **8** in moderate-to-good yields and with good to excellent diastereoselectivies. The sense of the stereo-induction was explained by the nonchelated transition-state model, which was consistent with the addition of organolithium reagents. The chiral auxiliary group on sulfinamide **8** could be readily removed by acidic alcoholysis to provide the corresponding trifluoroethylamine analog **9** (TABLE **4**) [27].

The application of chiral trifluoromethyl α , β unsaturated *N-tert*-butylsulfinyl ketimines in the asymmetric synthesis of trifluoromethylated



Figure 2. Different transition-state models.

allylic amines was demonstrated by Liu and coworkers [28]. Either diastereomer of the trifluoromethylated allylic amines was obtained in good yield and with excellent diastereoselectivity via the diastereoselective reduction of chiral trifluoromethyl α , β -unsaturated *N*-*tert*-butylsulfinyl ketimines **10** by diisobutylaluminium hydride (DIBAL-H) and L-selectride, respectively **(TABLE 5)** [28]. The reversal of the diastereofacial selectivity observed with different reducing agents could be explained by different transition states involved: a cyclic (closed) transition state in the reduction of DIBAL-H and an open transition state in the reduction of L-selectride (FIGURE 2) [28].

Chiral allylic primary amines **15** are potential dipeptide mimics, and their synthesis has been achieved through the efficient asymmetric reductive amination of fluoroenones **13 (TABLE 6)** [29]. Both diastereomers were obtained from the same chiral nonracemic sulfinyl imine by using different reducing agents. Moreover, neither the fluoroolefin geometry nor the presence of a chiral center on the C-terminal moiety affected the reduction selectivity [29]. The origins of different diastereoselectivities with different reducing agents (e.g., DIBAL-H and L-selectride) can be depicted with similar transition states, as shown in **FIGURE 2**.

The synthesis of chiral fluorinated sulfinamides 14 was also achieved via diastereoselective addition of Grignard and organolithium reagents to *N*-(*tert*-butylsulfinyl)- α -fluoroenimines 16 (TABLE 7) [30]. All organometallic reagents were added regioselectively at the imino carbon of α -fluoroenimines 16 to provide chiral allylamines in excellent yields and with diastereomeric ratios of up to 96:4. Enantiopure fluoroolefin dipeptide mimics could be obtained by acid-mediated removal of the sulfinyl group and simple chemical transformations.

Fluorinated chiral amines from nonfluorinated tert-butylsulfinimines & fluorinated reagents

The aforementioned asymmetric synthesis of fluorinated chiral amines was accomplished by using fluorinated *tert*-butylsulfinimines and their derivatives. However, most fluorinated *tert*butylsulfinimines and their derivatives were easily hydrolyzed or decomposed during work-up and/or storage, limiting their large-scale applications. Another useful strategy for asymmetric synthesis of fluorinated chiral amines has been based on the stereoselective nucleophilic addition of fluoroalkylation reagents to nonfluorinated *tert*-butylsulfinimines. This direct nucleophilic

FLUOROALKYLATION

A chemical reaction (or reaction sequence) that introduces one or more fluorine-bearing alkyl group(s) into substrates

Table 3. Diastereoselective 1,2-addition of aryllithium reagents to imine 5.						
H ₂ N ^{-S} +	$ \begin{array}{c} OH \\ F_3C \\ OH \\ Toluene \\ \end{array} \left[\begin{array}{c} OH \\ F_3C \\ F_3C \end{array} \right] $	° ° 5]-	RLi P THF, -78°C F ₃ C			
Entry	RLi	Product 7	Yield (%)	dr		
1	PhLi*	7a	66	98:2		
2	PhLi [‡]	7a	42	98:2		
3	PhLi [§]	7a	58	96:4		
4	4-MeOPhLi	7b	55	97:3		
5	4-MeSPhLi	7c	53	98:2		
6	4-FPhLi	7d	50	100:1		
7	3,5-diFPhLi [¶]	7e	36	83:17		
8	2-MePhLi	7f	40	99:1		
9	Pyridin-2-yllithium	7g	15	98:2		
*2.5 equiv of PhLi [‡] 1.1 equiv of PhLi [§] Precomplexation [¶] Unoptimized con Data from [23].	was used. was used. of imine with AIMe ₃ at 0°C, added to Pl ditions.	nLi solution at -78°C.				





Figure 3. Synthesis of α -monofluoromethylated cyclic secondary amine 32. Data from [44].

fluoroalkylation method was promoted by the development of a variety of fluoroalkylation reagents, such as Ruppert-Prakash reagent [(tri-fluoromethyl)trimethylsilane (TMSCF₃) and fluorinated sulfones, among others.

In 2001, Prakash and co-workers reported the first stereoselective synthesis of trifluoromethylated amines through nucleophilic trifluoromethylation of *N-tert*-butylsulfimines by using TMSCF₃ reagent (**TABLE 8**) [31]. Tetrabutylammonium triphenyldisilicate (TBAT) was applied as a hydrophobic fluoride initiator. The nucleophilic trifluoromethylation reaction can be applied to nonenolizable, enolizable, aromatic and heterocyclic imines alike, and the addition proceeded from the less hindered *Re* face of the imines to give the selective Cram products **18**, which can be transformed to the trifluoromethylated amines **19** [31].

TMSCF, is also an efficient nucleophilic trifluoromethylating reagent for the direct preparation of trifluoromethylated vinical ethylenediamines (TABLE 9) [32]. The nucleophilic trifluoromethylation of α-amino N-tert-butylsulfinimines was achieved with TMSCF, and afforded high yield of the trifluoromethylated vinical ethylenediamines in high stereoselectivity, particularly in the case of imines derived from L-amino acids. The very high diastereoselectivity observed in the reaction suggests that both chiral centers present in the molecule induce the incoming nucleophile to the *Re* face of the imines [32]. While in the case of imine derived from D-amino acid, despite the fact that each chiral center directed the addition from a different face of the imine, the major diastereomer was obtained from the Re face attack of imine, indicating that the tert-butylsulfinyl group controlled the diasterofacial selectivity of the reaction [32].

With TMSCF₃ reagent, Prakash and co-workers also developed the asymmetric synthesis of trifluoromethylated allylic amines by highly diastereoselective addition to α , β -unsaturated *N-tert*-butylsulfinimines by using TBAT or tetramethylammonium fluoride (TMAF) as a fluoride initiator **(TABLE 10)** [33]. In this reaction, there is no 1,4-addition product obtained and the rationale for the high anti-Ellman products could be explained by Cram-Davis's open transition-state model [33].

Fluorinated Reformatsky reagent was also found to readily react with *N-tert*-butylsulfinimines. Addition of the Reformatsky reagent **24** derived from ethyl bromodifluoroacetate to *N-tert*-butylsulfinimines produced β -branched α, α -difluoro- β -amino esters 25 in moderate-to-good diastereomeric ratios (dr 80:20 to 95:5) (TABLE II) [34]. The diastereomers were easily separated, and the resulting enantiopure protected β -amino esters were readily transformed to the corresponding acid, amide and amine derivatives as useful building blocks for medicinal chemistry applications.

Since 2003, difluoromethyl phenyl sulfone was extensively used as a versatile reagent for the selective transfer of difluoromethyl, difluoromethylene and difluoromethylidene building blocks [35-39]. In 2005, Li and Hu reported the first highly stereoselective and facile synthesis of α -difluoromethyl amines via nucleophilic difluoromethylation of N-(tert-butylsulfinyl)- aldimines with difluoromethyl-phenyl sulfone, followed by the convenient deprotection of both tert-butylsulfinyl and phenylsulfonyl groups (TABLE 12) [40]. The sense of diastereoselective induction can be depicted by a nonchelation-controlled addition step to give the Cram products 26.

In 2007, Li and Hu reported the diastereoselective (phenylthio) diffuoromethylation of N-(tertbutylsulfinyl)aldimines with the Me₂SiCF₂SPh reagent, which provides an alternative approach

Table 5. Diastereoselective reduction of 10 with DIBAL-H and Reducing agent HN THE F₃C 10 11 (Rs, R) 12 (Rs. S) R **DIBAL-H** L-selectride 11 yield (%) (dr) 12 yield (%) (dr) C₆H₅ 99 (99:1) 99 (99:1) 4-MeOC₆H₄ 92 (98:2) 98 (99:1) 4-MeC_cH₄ 97 (>99:1) 95 (98:2) 4-CIC₆H₄ 98 (99:1) 88 (99:1) 4-BrC₆H₄ 97 (99:1) 96 (98:2) 2-MeOC₂H₂ 97 (99:1) 96 (97:3) 3-BrC₆H₄ 90 (99:1) 93 (97:3) 1-Naphthyl 94 (99:1) 96 (99:1) 2-Furyl 94 (>98:2) 97 (98:2) C_cH_cC≡C 93 (> 96:4) 81 (95:5) 60 (96:4) n-C₀H₁₇ 87 (>99:1) $C_{e}H_{s}(CH_{s})_{s}$ 62 (99:1)

81 (>99:1)



Data from [28].

Figure 4. Diastereoselective monofluoromethylation of ketimines (35). Condition A: n-BuLi used as a base; condition B: KHMDS used as a base.

*Determined by ¹⁹F NMR using PhCF₃ as internal standard. Data from [46].

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Figure 5. Different transition-state models. Data from [46].



for the preparation of optically pure α -difluoromethyl amines (TABLE 13) [41]. Nucleophilic (phenylthio)difluoromethylation of *N*-(*tert*butylsulfinyl)imines afforded the corresponding products in good yields and with high diastereoselectivity. The PhSCF₂-containing sulfinamides obtained could be further transformed into chiral 2,4-*trans*-disubstituted 3,3-difluoropyrrolidines through an intramolecular radical cyclization methodology [41].

 α -monofluoromethyl amines have been extensively used as important building blocks in the design of many anticholinergic, antiemetric, antispastic drugs and enzyme inhibitors, given the fact that the fluorine atom lowers the basicity of amine functionality, decreases acute toxicity and increases the metabolic stability of a target drug [42,43]. The unprecedented nucleophilic monofluoromethylation of N-(tert-butylsulfinyl)imines using fluoromethyl phenyl sulfone was shown to be a highly stereoselective and convenient method for the synthesis of enantiopure α -monofluoromethyl amines (TABLE 14) [44]. The same methodology could also be used to synthesize homochiral α-monofluoromethylated cyclic secondary amine 32 by using tosylate-bearing N-tertbutylsulfinylimine 31 (FIGURE 3) [44].

PhSO₂CF₂H and PhSO₂CH₂F were robust nucleophilic di- and mono-fluoromethylating agents, the application of which was extended to the highly diastereoselective synthesis of homochiral α -diffuoromethylated and α -monofluoromethylated vicinal ethylenediamines through direct nucleophilic di- and monofluoromethylation of α -amino N-tert-butylsulfinimines 20 (TABLES 15 & 16) [45]. The excellent diastereoselectivity observed in the fluoroalkylation of imines 20 derived from the corresponding (S)-amino aldehvde indicated that both the tert-butylsulfinyl group and the dibenzylamino group induced the nucleophilic addition from the Re face of imines and the stereochemistry of the reactions was nonchelation controlled. The difluoromethylation of imine 20 derived from (R)-amino aldehyde gave the desired product with 95:5 diastereomeric ratio, which suggests that the tert-butylsulfinyl group (instead of the dibenzylamino group) dominated the stereochemical outcome of the addition products. Interestingly, moderate stereoselectivies (1.5:1 to 3.3:1) were observed in the case of monofluoromethylation of imines 20 during the formation of another neighboring stereogenic center (the fluorine-bearing carbon) (TABLES 15 & 16) [45].

Table 7. Diastereoselective	addition of c	rganom	etallic rea	gents to α -	fluoroenir	nines 16.
$R_1 \xrightarrow{P}_{F} H$	ene H		+ R1			
16	(<i>Ss</i> ,	, <i>S</i>)-14	(<i>Ss</i> ,	<i>R</i>)-14		
Substrate 16	R ₂ M	T (°C)	Time (h)	Additive	Yield (%)	dr
ļ Ģ	PhMgBr	-30	2		95	10:90
N ^Ś	PhLi	-78	3		95	67:33
	MeMgBr	0	0.75	AlMe ₃	82	6:94
H I I I I I I I I I I I I I I I I I I I	MeLi	-78	5		96	45:55
MeO	i-PrMgCl	0	0.7		60	30:70
	i-PrLi	-78	1		98	60:40
	i-BuMgBr	0	0.7		27	30:70
	BnMgCl	0	1		94	30:70
	AllyMgBr	0	1		82	4:96
	VinylMgBr	0	0.5		95	33:67
Q Q	PhMgBr	0	1.5		90	14:86
N ^Ś	PhLi	-78	12		83	65:35
	MeMgBr	0	2.5	AlMe ₃	84	10:90
TBDPSO' Y H	i-PrMgCl	0	0.75		41	49:51
F	i-PrLi	-78	1		91	67:33
	i-BuMgBr	0	0.5		36	30:70
	BnMgCl	0	1.5		90	43:57
	AllyMgBr	0	0.75		91	8:92
	VinylMgBr	0	0.75		98	35:65
Data from [30].						



A: Open TS (in THF) H R B: Closed TS(in toluene) $G = -CF_3$ F_3C

Figure 6. Different transition-state models. Data from [47].





Table 8. Nucleophilic trifluorom	ethylation of sulfinylimines 1	17.
$\begin{array}{ c c } & & & \\ & & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & &$	$ \begin{array}{c} $	F ₃ `R
R	Yield (%)	dr
Ph	80	97:3
4-BrC ₆ H ₄	90	>99:1
4-CIC ₆ H ₄	95	>99:1
4-CF ₃ C ₆ H ₄	84	95:5
2-pyridyl	95	99:1
3-pyridyl	92	99:1
2-furyl	85	97:3
2-naphthyl	83	96:4
9-anthryl	90	99:1
Cyclohexyl	88	99:1
<i>t</i> -Bu	75	99:1
CH ₂ CH ₂ Ph	84	90:10
Data from [31].		



Data from [32].

Table 10. Nucleophilic	trifluoromethylation of $lpha,eta$ -unsati	urated <i>N-tert</i> -butylsulfinimines 22.
$\begin{array}{c} 0 H R^2 \\ \hline S N f R^3 - \\ 22 R^1 T T R^3 - \\ \end{array}$	$\xrightarrow{\text{TMSCF}_3} \xrightarrow{\text{O}} \overset{\text{CF}_3}{\text{S}} \overset{\text{R}^1}{\underset{\text{H}}{\overset{\text{CF}_3}}} \overset{\text{R}^1}{\underset{\text{R}^3}{\overset{\text{CF}_3}{\underset{\text{H}}{\overset{\text{R}^3}}}} \overset{\text{R}^3}{\underset{\text{R}^3}{\overset{\text{CF}_3}{\underset{\text{R}^3}}}}$	
Product 23	Yield (%)	dr
$\begin{array}{c} O & CF_3 H \\ \hline S & \downarrow \\ H & H \end{array}$	55	90:10*
S N H Me	73	>99:1*
S N H H H	76	92:8*
S N H Me	50	98:2*
$\begin{array}{c} O & CF_3 & Ph \\ \hline S & H & H \\ H & H \end{array}$	62	>99:1*
$\begin{array}{c} O \underbrace{CF_3 Ph}_{F_3} \\ S_{N} \\ H \\ (CH_2)_4CH_3 \end{array}$	82	90:10 [‡]
$\begin{array}{c} O \underset{{{}{{}{}{}{}{\overset$	75	92:8 [‡]
$\begin{array}{c} O CF_3 Ph \\ \hline \\ S \\ H \\ Ph \end{array}$	62	93:7 [‡]
*TBAT as the fluoride source. *TMAF as the fluoride source. Data from [33].		

Table 11. Diastereoselective	addition of 24 to chira	al sulfinimines 17	
$ \xrightarrow{\begin{array}{c} 0 \\ S \\ N \\ 17 \end{array}} \xrightarrow{\begin{array}{c} BrZn \\ F \\ F \\ F \\ THF \end{array}} \xrightarrow{\begin{array}{c} CO_2 Et \\ F \\ CO_2 Et \\ F \\ THF \end{array}} $			
Substrate	Product	Yield (%)	dr
$R = i - C_4 H_9 (17a)$	25a	51	81:19
$R = n - C_3 H_7 (17b)$	25b	55	80:20
R = Ph (17c)	25c	82	90:10
$R = c - C_6 H_{11} (17d)$	25d	81	87:13
R = 2-thiazolyl (17e)	25e	58	95:5
Data from [34].			

Table 12. Stereoselective (ph	enylsulfonyl)difluor	omethylation of chiral sulfinylimines 17.
$\begin{array}{c} 0 \\ S \\ N \\ R \\ R \\ T \\ 17 \end{array}^{+ PhSO_2CF_2H} - T$	LHMDS HF, -78 °C	$ \begin{array}{c} {}_{2}SO_{2}Ph & CF_{2}H \\ R & \xrightarrow{} \overline{C}I H_{3}N \\ 99\% ee \\ 27 \end{array} $
R	Yield (%)	dr
Ph	95	>99:1
4-MeOC6H4	96	>99:1
4-CIC6H4	95	>99:1
2-naphthyl	98	>99:1
2-furyl	90	>99:1
Et	95	>99:1
i-Pr	94	>99:1
t-Bu	85	>99:1
Data from [40].		

 Table 13. Stereoselective (phenylthio)difluoromethylation of sulfinylimines 17.

$\begin{array}{c} O \\ S \\ N \\ R \end{array} \xrightarrow{TMSCF_2SPh, TBAT} \\ DMF, -40 \text{ to } -20^{\circ}C \end{array}$	$\begin{array}{c} O \\ S \\ N \\ H \\ 28 \end{array} \xrightarrow{F} \\ R \\ $	
R	Yield (%)	dr
Ph	75	≥99:1
4-MeOC6H4	85	≥98:2
4-CIC6H4	89	≥99:1
2-naphthyl	85	≥98:2
2-furyl	72	≥98:2
iPr	75	≥99:1
c-C6H11	74	≥99:1
n-Pr	30	≥99:1
t-Bu	71	≥99:1
YYY NBn₂	58	≥99:1
Data from [41].		

Table 14. Facile synthesis of lpha-monofluoromethyl amines 30.

$\begin{array}{c} 0 \\ S \\ N \\ 17 \end{array}$	a) PhSO ₂ CH ₂ F, LHMDS, THF b) Na(Hg), Na ₂ HPO ₄ , MeOH c) HCI (dioxane), MeOH	• FH ₂ C 30	
R	Yield (%)		dr
Ph	77		99:1
4-MeOC ₆ H ₄	76		99:1
4-CIC ₆ H ₄	73		99:1
2-naphthyl	74		99:1
4-Me ₂ NC ₆ H ₄	75		98:2
2-furyl	71		99:1
t-Bu	73		99:1
i-Pr	70		99:1
n-Pr	70		99:1
CH,CHMe,	74		99:1
Data from [44].			

Table 15. Difluoromethyla	ation of α-amino <i>N-tert</i> -k	outylsulfinimines 20.
O tBu ^S N ^R NBn ₂ R	PhSO ₂ CF ₂ H aHMDS, THF -78°C	$ \begin{array}{c} CF_2SO_2Ph\\ \vdots\\NBn_2\\R\\33\\33\\SO_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\NBn_2\\\mathsf$
Product	Yield (%)	dr
$t_{Bu} \xrightarrow{O}_{H} \underbrace{\underset{CH_2}{CF_2}}_{S_{V}} \underbrace{\underset{CH_2}{CPh}}_{CH_2} N Bn_2$	80	>99:1
$^{P}_{\text{Bu}} \xrightarrow{S} \overset{CF_2SO_2Ph}{\overbrace{,}} \overset{NBn_2}{\underset{H}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{VBn_2}{\overset{V}}{\overset{V}{\overset{V}{\overset{V}}{\overset{V}{\overset{V}}{\overset{V}{\overset{V}}{\overset{V}}{\overset{V}{\overset{V}{\overset{V}}{\overset{V}{\overset{V}}{\overset{V}{\overset{V}}{\overset{V}}{\overset{V}{\overset{V}}{\overset{V}}{\overset{V}}{\overset{V}}{\overset{V}{\overset{V}}{\overset{V}}{\overset{V}{\overset{V}}}}}}}}$	83	>99:1
$^{O}_{Bu}$ $\overset{CF_{2}SO_{2}Ph}{\overset{T}{}_{}}_{H}$ $\overset{NBn_{2}}{}_{CH(CH_{3})_{2}}$	74	>99:1
$^{P}Bu \xrightarrow{S} N \xrightarrow{CF_2SO_2Ph}_{H} H \xrightarrow{CH_2Ph}_{CH_2Ph}$	85	95:5
$^{\prime}Bu \xrightarrow{CF_2SO_2Ph}_{H} \overset{CF_2SO_2Ph}{\xrightarrow{Ph}}_{Ph}$	73	>99:1
^O ^{CF₂SO₂Ph ⁱBu^SN^NH^{CH}(CH₃)CH₂CH}	83	>99:1
$^{\prime}Bu \xrightarrow{S} N \xrightarrow{CF_2SO_2Ph}_{H} H \xrightarrow{CH_3} NBn_2$	95	>99:1
Data from [45].		

Although the stereoselective fluoroalkylations of N-tert-butylsulfinyl aldimines have been successfully accomplished to afford chiral α -fluoroalkyl amines (as mentioned above), the efficient and stereoselective synthesis of α -fluoroalkyl tertiary carbinamines by nucleophilic fluoroalkylation of ketimines is more difficult. The first highly diastereoselective synthesis of α , α -dibranched monofluoromethyl amines was achieved by Hu and co-workers via monofluoromethylation of N-tert-butylsulfinyl ketimines using the pregenerated fluoro(phenylsulfonyl)methyl anion (FIGURE 4) [46]. The pregeneration of PhSO₂CHF⁻ anion from the PhSO₂CH₂F and a base played a key role in the reaction, and the thermal stability, good nucleophilicity and relatively weak basicity of the PhSO₂CHF⁻ anion accounted for the overall chemical outcome of the reaction. The stereocontrol mode of the current diastereoselective monofluoromethylation of ketimines was opposite to the other known nucleophilic fluoroalkylation of *N-tert*-butylsulfinyl aldimines (FIGURE 5A), which suggests that a cyclic six-membered (closed) transition state is involved in the reaction (FIGURE 5B) [46].

N-tert-butylsulfinimines were also subjected to the 1,2-additions of trifluoromethylacetylide (prepared from lithium diisopropylamide [LDA] and the 2-bromo-3,3,3-trifluoropropene, *in situ*) to give homochiral trifluoromethylated propargylamines [47]. In the case of *N-tert*-butylsulfinyl aldimines (TABLE 17), the diastereoselectivity



could be tuned by the use of a polar or nonpolar solvent (FIGURE 6); while the addition of trifluoromethylacetylide to *N-tert*-butylsulfinyl ketimines and the subsequent significant improvement of the diastereoselectivity was achieved through the use of AlMe, as an adductive (TABLE 18) [47].

The bis(trifluoromethylation) of imine **40** with TMSCF₃ (using TMAF as initiator) was also found to be successful, and both trifluoromethyl groups were added to the *Si* face of the imine *E*-bonds, giving (S_s,S_s) -N,N'-(1R,1R')-1,1'-(anthracene-9,10-diyl)-bis(2,2,2-trifluoroethane-1,1-diyl)-bis(2-methylpropane-2-sulfinamide) **(41)** as a single isomer **(FIGURE 7)** [48]. Hydrolysis of **41** followed by neutralization afforded bis(trifluoromethylated) diamine **42**.

The electron-withdrawing trifluoromethyl group greatly reduces the donating ability of the electron lone pairs on the nitrogen atoms, thus decreasing the basicity of the amine and increasing the acidity of the corresponding ammonium salts.

Future perspective

Significant progress has been made in the development of stereocontrolled synthesis of fluorinated chiral amines, the majority of which have been accomplished based on chiral auxiliaries, especially on Ellman's N-tert-butylsulfinyl imines. These synthetic methodologies provide a simple and efficient way to get access to a variety of structurally diverse fluorinated chiral amines, which are highly useful for developments in medicinal chemistry. As discussed above, fluorinated chiral amines have been designed as potential peptidomimetics (TABLE 6) [29], tripeptides [34] and 2-oxa-1-imidazolidinyl derivatives [32], among others [1–8]. It should be noted that, although catalytic enantioselective monofluoromethylation of in situ-generated imines (from α -amido sulfones) under the combination of fluorobis(phenylsulfonyl)methane chemistry with a Mannich-type reaction was recently reported [49], the enantioselective synthesis of tri- and di-fluoromethylated chiral amines via nucleophilic fluoroalkylation still remains a challenging task. Considerable attention should be focused on the organocatalytic or metallocatalytic enantioselective synthesis of fluorinated chiral amines in the coming years.

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Table 18. 1,2-Addition of lithium trifluoromethylacetylide to

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ketimines 35.			
O Me S N R − 35	$Li - CF_3$ Me_3AI , toluene	V Me R S N H CF_3 39	
R	Yield (%)	dr	
iPr	88	94:6	
iBu	83	93:7	
Et	69	>99:1	
Ph	31	>99:1	
Data from [47].			

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Executive summary

patents received or pending, or royalties.

- The stereoselective synthesis of fluorinated chiral amines based on the use of Ellman's N-tert-butylsulfinyl imines has been accomplished by two major strategies.
- One is the stereoselective addition or asymmetric reduction of fluorinated *tert*-butylsulfinimines and their derivatives. The unstability of fluorinated *tert*-butylsulfinimines limits the scope of their applications.
- The other strategy is the asymmetric addition of fluorinated reagents to *tert*-butylsulfinyl imines, the development of which is lacking new efficent and atom-economic fluorinated reagents.
- These highly useful synthetic methodologies provide powerful tools for the synthesis of optically pure fluorinated amines, amino acids and peptides, and promise to find many applications in the future for medicinal chemistry.

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Following the publication of the Mini-Review by Jinbo Hu et al. 'Synthesis of fluorinated chiral amines using N-tert-butylsulfinyl imines, published in the August 2009 issue of Future Medicinal Chemistry (Future Med. Chem. 1[5], 875–888 [2009]), it has been brought to our attention that:

• The footnote of Table I was incorrectly printed as: TMSCN: Trimethyl cyanide. Data from [15]. This should have read: TMSCN: Trimethylsilyl cyanide. Data from [21].

• A sentence on page 877 was incorrectly printed as:

"Chiral allylic primary amines **I5** are potential dipeptide mimics, and their synthesis has been achieved through the efficient asymmetric reductive amination of fluoroenones **I3** (FIGURE 2) [29]."

This should have read:

"Chiral allylic primary amines **I5** are potential dipeptide mimics, and their synthesis has been achieved through the efficient asymmetric reductive amination of fluoroenones **I3** (TABLE 6) [29]."

• Table 3 was incorrectly printed as:

Table	3. Diastereoselective 1,2-addi	tion of aryllithium	reagents to imine	e 5 .
	H_2N F_3C H_2N H_2N H_3C	$\begin{bmatrix} 0 \\ F_{3}C & N \\ 5 \end{bmatrix} = \begin{bmatrix} 0 \\ -1 \\ -1 \\ -1 \\ -1 \\ -1 \\ -1 \\ -1 \\ $	RLi THF, -78°C F ₃ C	Ç S
Entry	RLi	Product 7	Yield (%)	dr
1	RLi	7a	66	98:2
2	PhLi*	7a	42	98:2
3	PhLi [‡]	7a	58	96:4
4	4-MeOPhLi	7b	55	97:3
5	4-MeSPhLi	7c	53	98:2
6	4-FPhLi	7d	50	100:1
7	3,5-diFPhLi§	7e	36	83:17
8	2-MePhLi	7f	40	99:1
9	Pyridin-2-yllithium	7g	15	98:2
*2.5 eau	iv of PhLi was used			

[±]1.1 equiv of PhLi was used.

+1.1 equiv of PhLi was used.

[§]Precomplexation of imine with AIMe₃ at 0°C, added to PhLi solution at -78°C. Unoptimized conditions. Data from [23].

This should have read:

Table 3. Diastereoselective 1,2-addition of aryllithium reagents to imine 5.				
H ₂ N	OH Nolecular sieves OH Toluene		RLi IF, -78°C F ₃ C N H 7	o s
Entry	RLi	Product 7	Yield (%)	dr
1	PhLi*	7a	66	98:2
2	PhLi [‡]	7a	42	98:2
3	PhLi [§]	7a	58	96:4
4	4-MeOPhLi	7b	55	97:3
5	4-MeSPhLi	7с	53	98:2
6	4-FPhLi	7d	50	100:1
7	3,5-diFPhLi ¹	7e	36	83:17
8	2-MePhLi	7f	40	99:1
9	Pyridin-2-yllithium	7g	15	98:2
[*] 2.5 equiv of PhLi was used. [‡] 1.1 equiv of PhLi was used. [§] Precomplexation of imine with AIMe ₃ at 0°C, added to PhLi solution at -78°C. [¶] Unoptimized conditions. Data from [23].				





This should have read:



• Table 5 row 10 was incorrectly printed as:

Table 5. Diastereoselective reduction of 10 with DIBAL-H and ı-selectride.			
R	DIBAL-H 11 yield (%) (dr)	∟-selectride 12 yield (%) (dr)	
	93 (> 96:4)	81 (95:5)	
Data from [21].			

This should have read:

Table 5. Diastereoselective reduction of 10 with DIBAL-H and <code>L-selectride.</code>			
R	DIBAL-H 11 yield (%) (dr)	∟-selectride 12 yield (%) (dr)	
C ₆ H ₅ C≡C	93 (> 96:4)	81 (95:5)	
Data from [28].			

• The footnote of Figure 4 should have read: "Figure 4. Diastereoselective monofluoromethylation of ketimines (35). Condition A: n-BuLi used as a base; condition B: KHMDS used as a base. *Determined by ¹⁹F NMR using PhCF₃ as internal standard. Data from [46]."

• A sentence on page 885-886 was incorrectly printed as:

"In the case of *N-tert*-butylsulfinyl aldimines, the diastereoselectivity could be tuned by the use of a polar or nonpolar solvent **(Figures 6 & 7)**; while the addition of trifluoromethylacetylide to *N-tert*-butylsulfinyl ketimines and the subsequent significant improvement of the diastereoselectivity was achieved through the use of AlMe₂ as an adductive **(TABLE 18)** [47]."

This should have read:

"In the case of *N-tert*-butylsulfinyl aldimines (TABLE 17), the diastereoselectivity could be tuned by the use of a polar or nonpolar solvent (FIGURE 6); while the addition of trifluoromethylacetylide to *N-tert*-butylsulfinyl ketimines and the subsequent significant improvement of the diastereoselectivity was achieved through the use of AlMe₃ as an adductive (TABLE 18) [47]."

• Table 13 was incorrectly printed. It should have read:

Table 13. Stereoselective (phenylthio)difluoromethylation of sulfinylimines 17.				
P H S N R TMSCF ₂ SPh, TBAT DMF, -40 to -20°C	$\begin{array}{c} O CF_2SPh \\ S N R \\ H \\ 28 \end{array} \xrightarrow{F} \begin{array}{c} F \\ F \\ R \\ N \\ H \\ R \\ H \\ H \\ R \\ H \\ R \\ H \\ R \\ R \\ H \\ R \\ H \\ R \\ $	κ ⁰		
R	Yield (%)	dr		
Ph	75	≥99:1		
4-MeOC6H4	85	≥98:2		
4-CIC6H4	89	≥99:1		
2-naphthyl	85	≥98:2		
2-furyl	72	≥98:2		
iPr	75	≥99:1		
c-C6H11	74	≥99:1		
n-Pr	30	≥99:1		
t-Bu	71	≥99:1		
YY NBn₂	58	≥99:1		
Data from [41].				

The authors and editors of *Future Medicinal Chemistry* would like to sincerely apologize for any inconvenience or confusion this may have caused our readers.