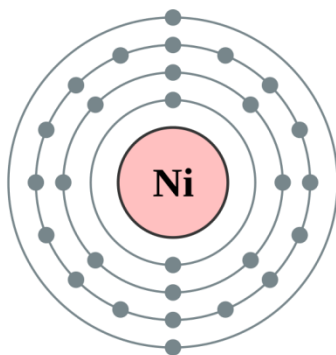


Recent Advances in Homogeneous Nickel Catalysis



Wenjun Miao

Oct. 27th 2014

1 Introduction

2 Cross-coupling Reaction

3 Heck Reaction

4 Reductive Coupling Reaction

5 Summary and Outlook

Basic Characteristics of Nickel

Nickel



-1 0 +1 +2 +3 +4

Electronegativity 1.8

Small atomic radius

Less electronegative

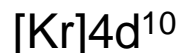
Harder

Facile Oxidation addition

Facile β -migratory insertion

Radical pathways more accessible

Palladium



0 +1 +2 +3 +4

Electronegativity 2.2

Larger atomic radius

More electronegative

Softer

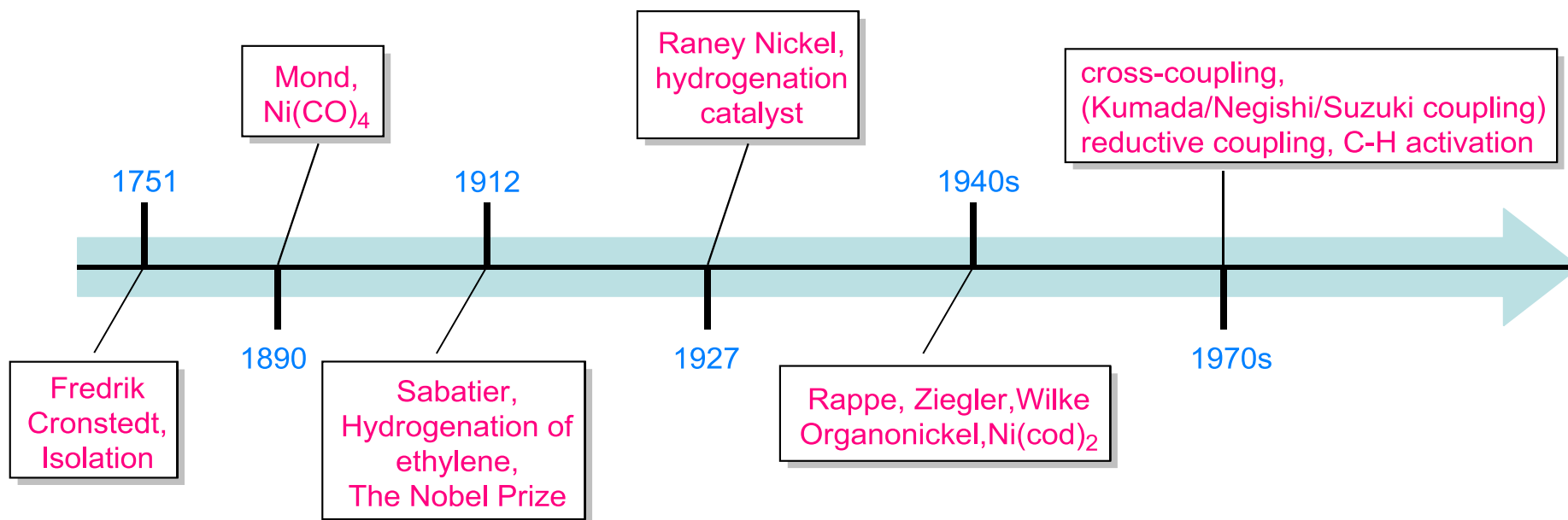
Facile reduction elimination

Facile β -hydride elimination

Price: The cost of Ni in its elemental form is roughly **2,000** times lower than **Pd** and **10,000** times lower than **Pt** on a mole-for-mole basis.

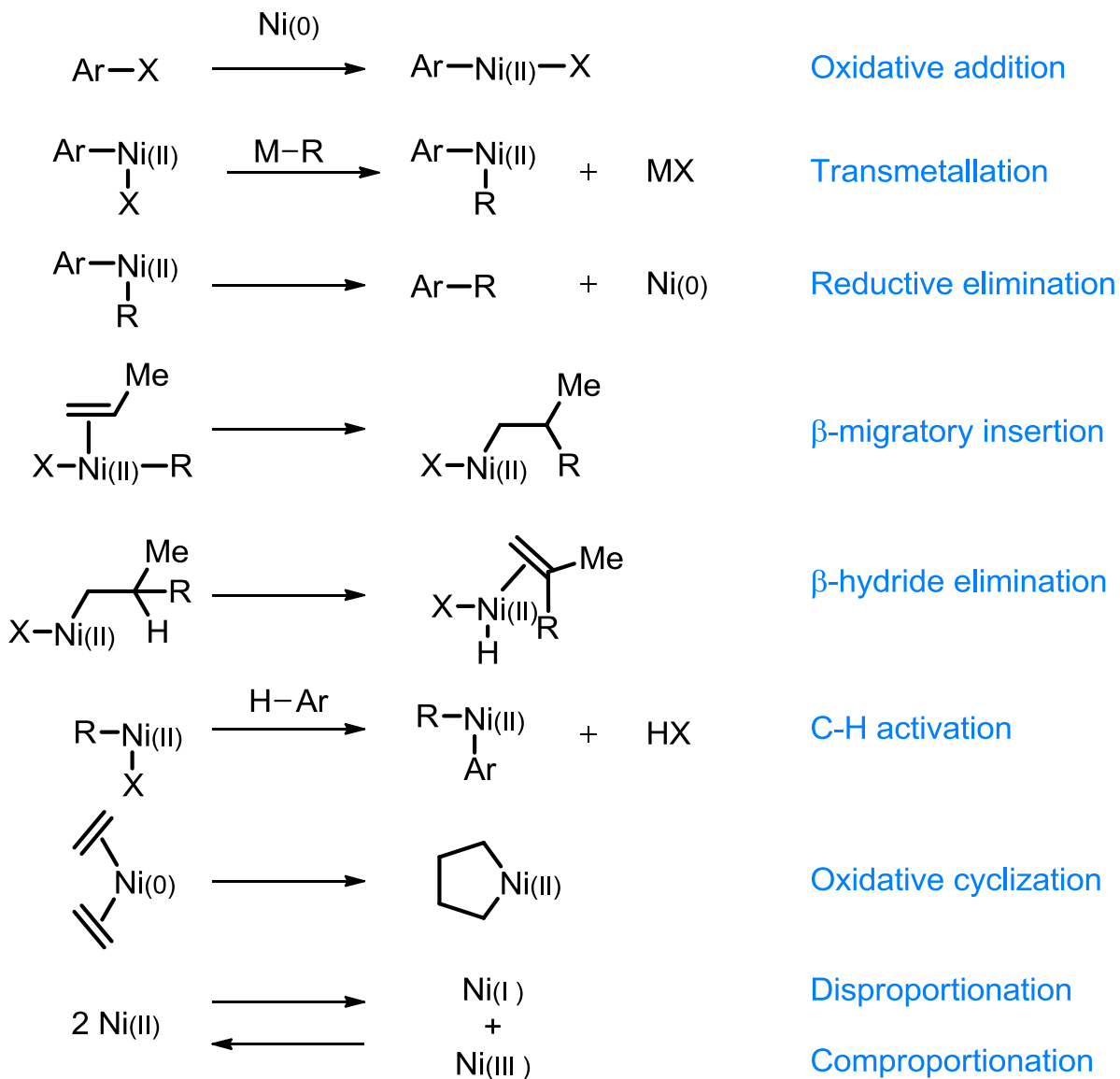
Disadvantage: Toxicity

Timeline of Achievements in Nickel Chemistry



Wilke, G. *Angew. Chem., Int. Ed.* **1988**, 27, 185.
Modern Organonickel Chemistry; Tamaru Y., Ed.; Wiley-VCH: Weinheim, 2005.

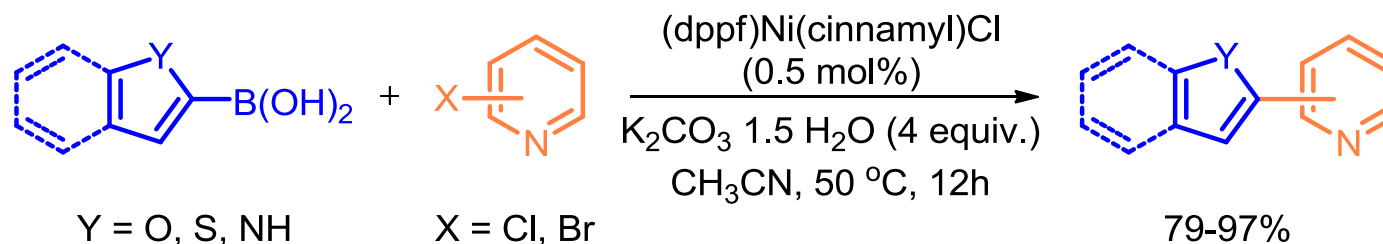
The Elementary Organometallic Reactions of Ni



Cross-coupling of Aryl Halides

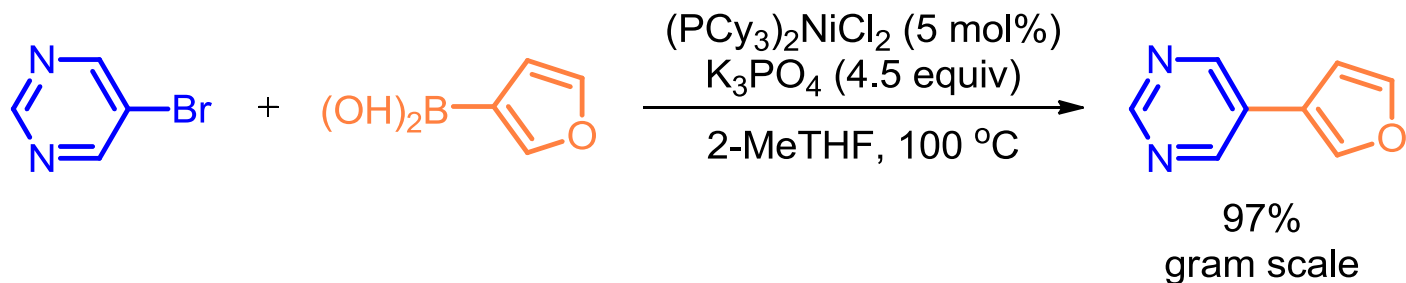
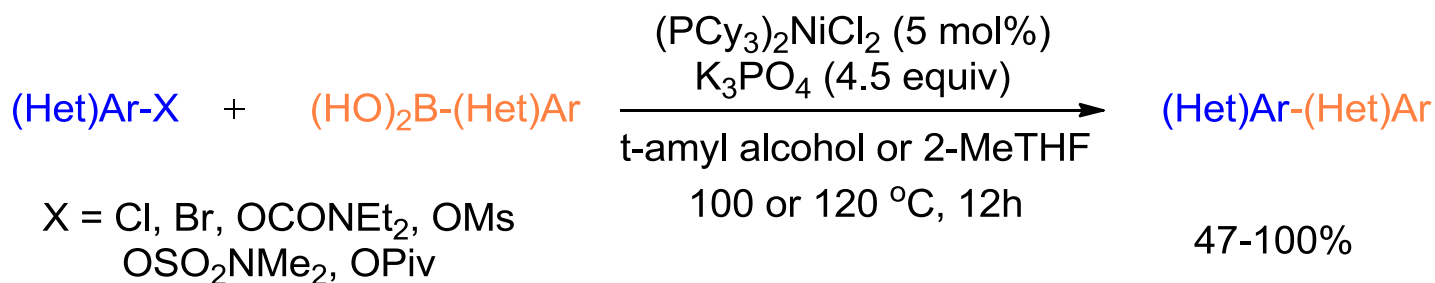
Challenges of catalytic synthesis of the hetero-biaryl compounds:

- The ligation of the heteroaryl coupling partner can poison the catalyst.
- The scope is limited .
- Heteroaromatic boronic acid: undergo protodeboronation under the basic reaction conditions.



Ge, S.; Hartwig, J. F. *Angew. Chem. Int. Ed.* **2012**, *51*, 12837.

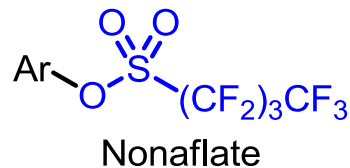
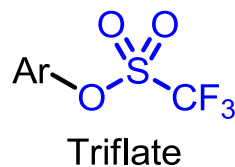
Cross-coupling of Aryl Halides



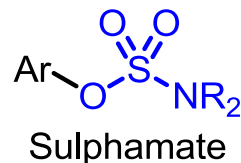
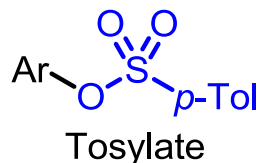
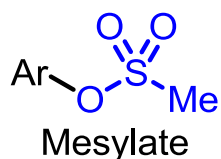
Ramgren, S. D.; Hie, L.; Ye, Y.; Garg, N. K. *Org Lett.* **2013**, *15*, 3950.

Cross-coupling of Phenol Derivatives

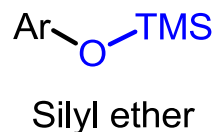
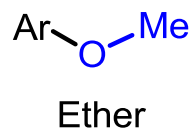
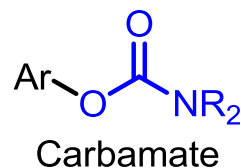
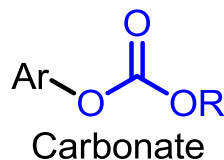
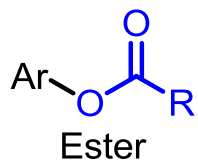
a



b



c

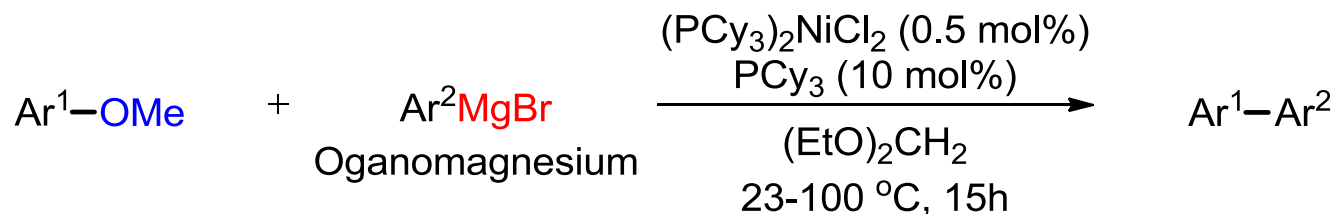


Ease of oxidation
addition by nickel

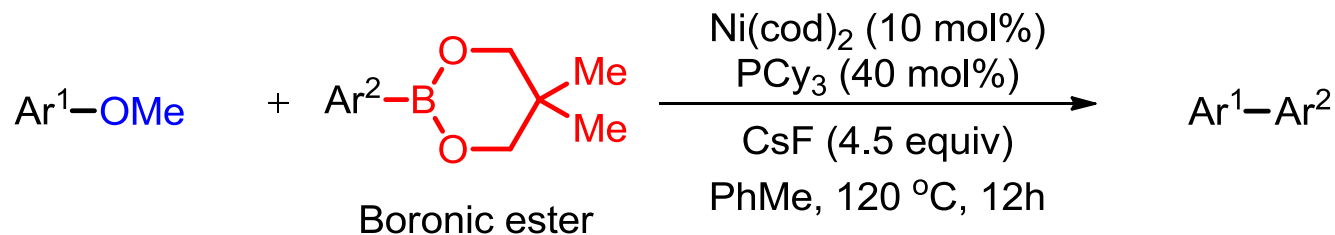
Triflate: Prone to hydrolysis, especially under basic conditions.

Tosylate, mesylate: Stable

Cross-coupling of Phenol Derivatives

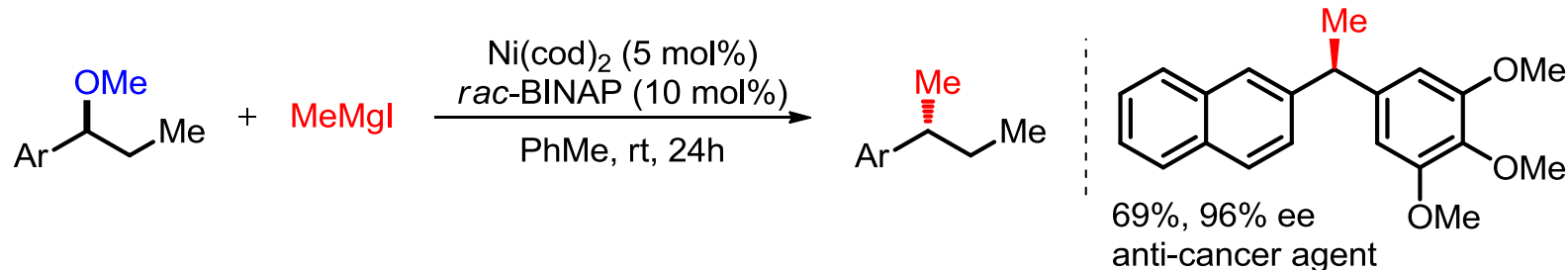


Dankwardt, J. W. *Angew. Chem. Int. Ed.* **2004**, *43*, 2428.



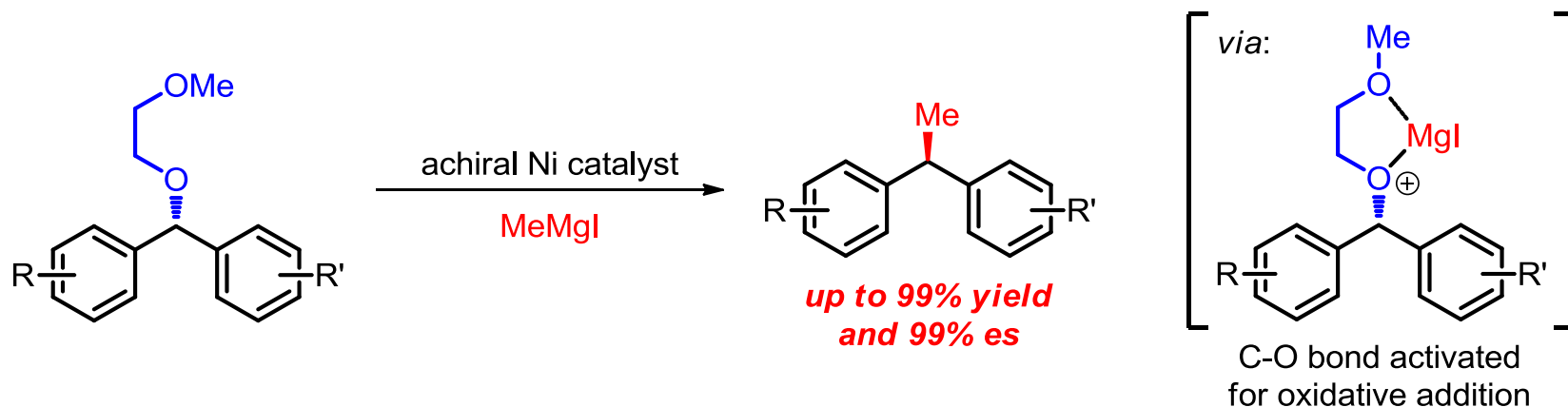
Tobisu, M.; Shimasaki, T.; Chatani, N. *Angew. Chem. Int. Ed.* **2008**, *47*, 4866.

Benzylic Cross-coupling



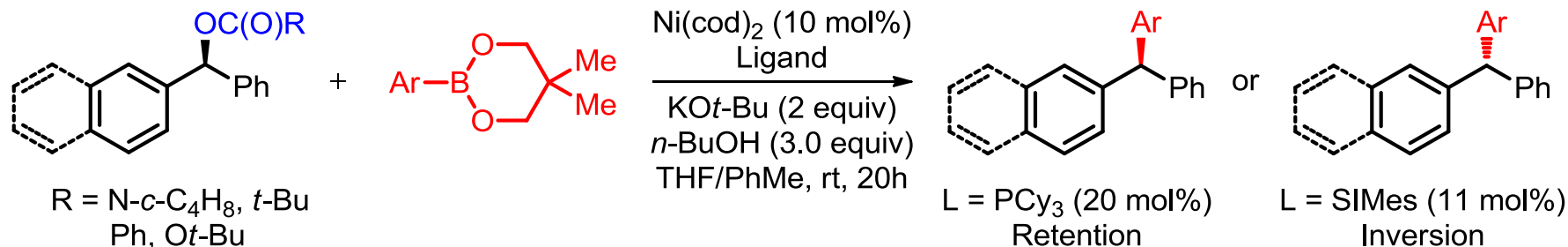
First stereospecific nickel-catalysed
alkyl-alkyl cross-coupling reaction

Taylor, B. L. H.; Swift, E. C.; Waetzig, J. D.; Jarvo, E. R. *J. Am. Chem. Soc.* **2010**, *133*, 389.

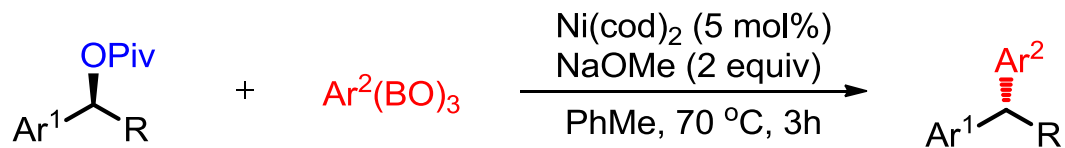


Greene, M. A.; Yonova, I. M.; Williams, F. J.; Jarvo, E. R. *Org Lett.* **2012**, *14*, 4293.

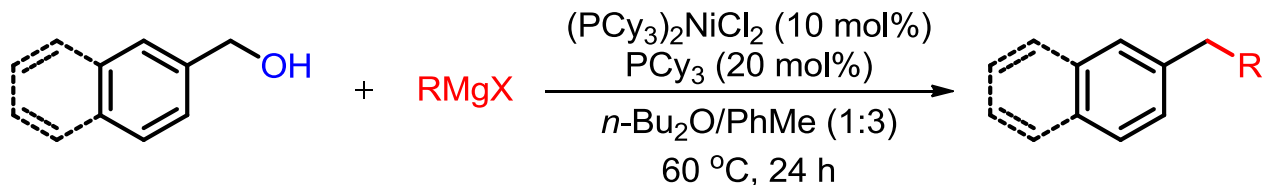
Benzylic Cross-coupling



Harris, M. R.; Hanna, L. E.; Greene, M. A.; Moore, C. E.; Jarvo, E. R. **2013**, *135*, 3303.

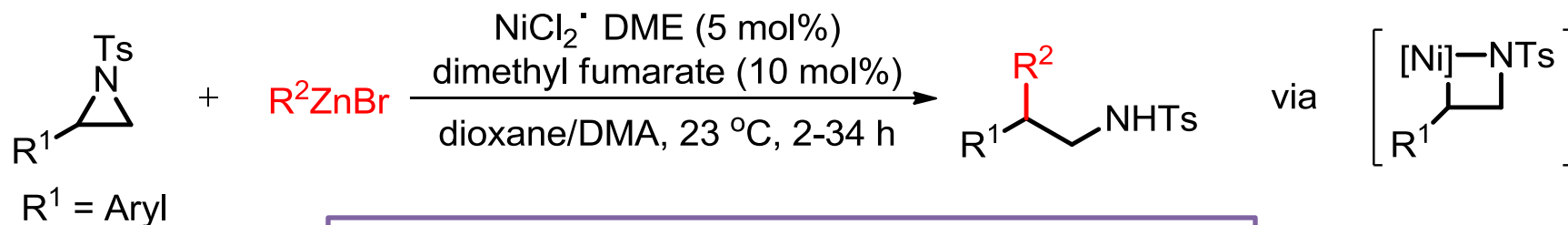


Zhou, Q.; Srinivas, H. D.; Dasgupta, S.; Watson, M. P. *J. Am. Chem. Soc.* **2013**, *135*, 3307.



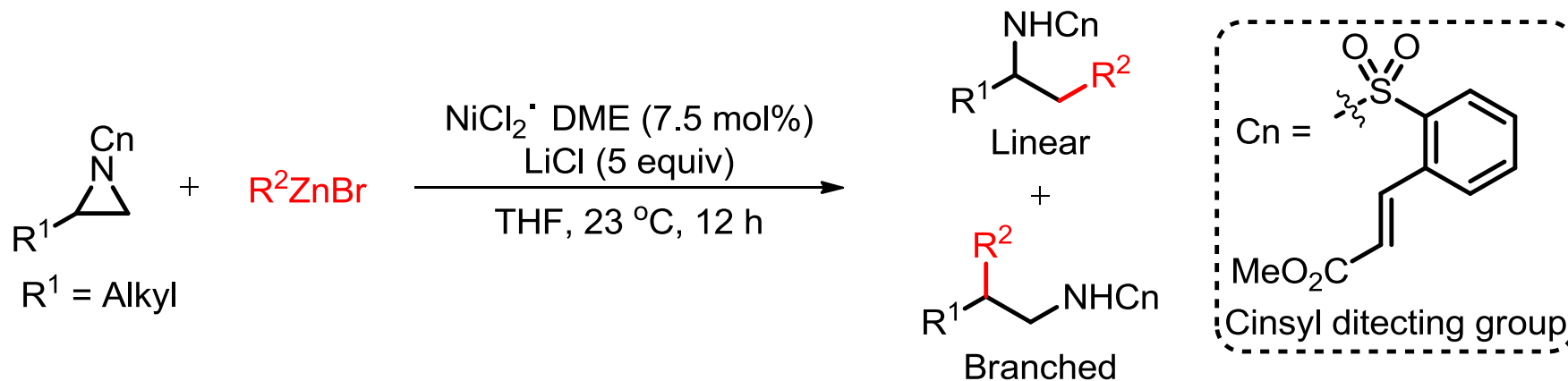
Yu, D.-G.; Wang, X.; Zhu, R.-Y.; Luo, S.; Zhang, X.-B.; Wang, B.-Q.; Wang, L.; Shi, Z.-J. *J. Am. Chem. Soc.* **2012**, *134*, 14638.

Cross-coupling of Aziridines



Dimethyl fumarate: accelerate reductive elimination through π -coordination to nickel

Huang, C.-Y.; Doyle, A. G. *J. Am. Chem. Soc.* **2012**, *134*, 9541.



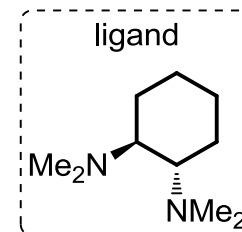
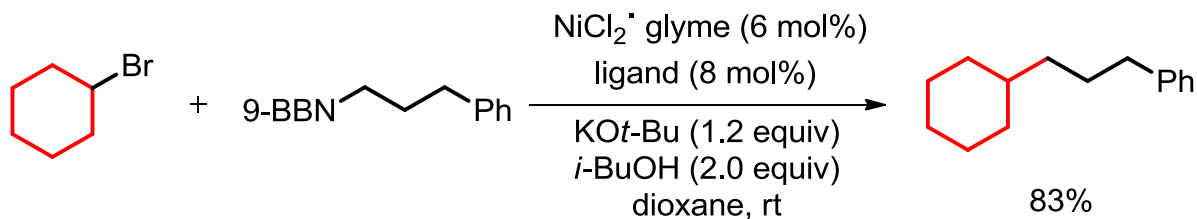
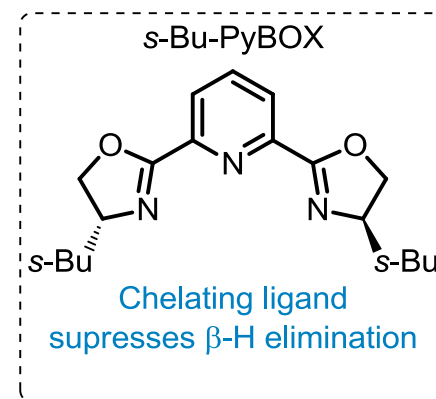
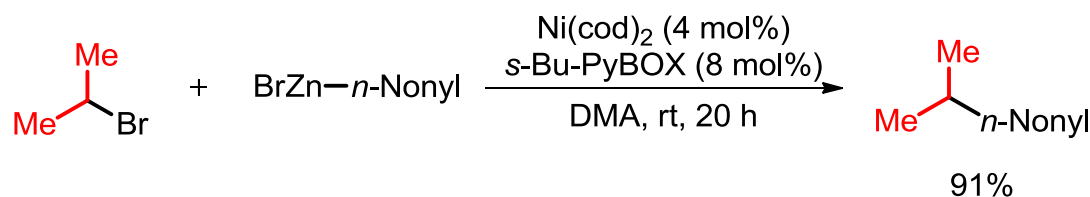
Nielsen, D. K.; Huang, C.-Y.; Doyle, A. G. *J. Am. Chem. Soc.* **2013**, *135*, 13605.

Cross-coupling of sp^3 Halides

Challenges:

◆ The electrophile is sp^3 hybridized, the activation energy for oxidative addition can be large, $C(sp^3)-X$ are more electron-rich than $C(sp^2)-X$, which can be reluctant to oxidatively add to the metal catalyst.

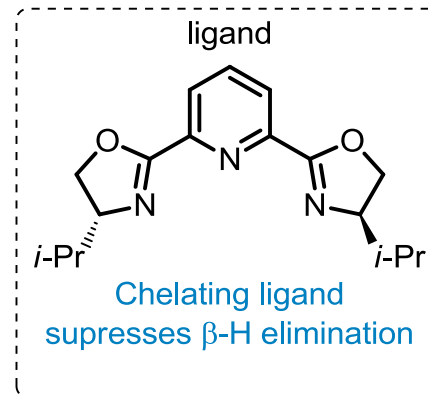
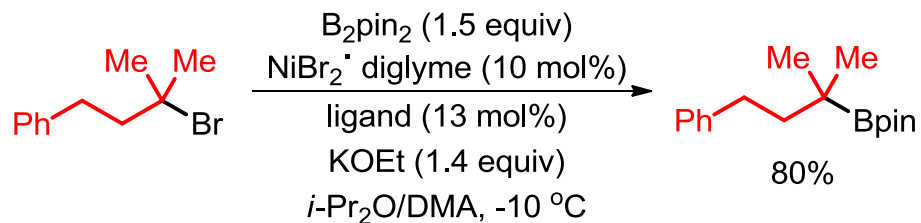
◆ Intramolecular β -H elimination



Zhou, J.; Fu, G. C. *J. Am. Chem. Soc.* **2003**, *125*, 14726.

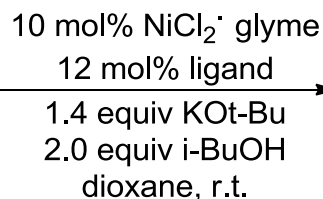
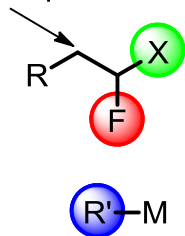
Zhou, J.; Fu, G. C. *J. Am. Chem. Soc.* **2004**, *126*, 1340.

Cross-coupling of sp^3 Halides



Dudnik, A. S.; Fu, G. C *J. Am. Chem. Soc.* **2012**, *134*, 10693.

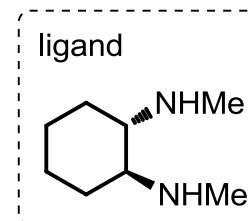
no functional group attached



X = I, Br, Cl
R' = Aryl, alkyl
R'-M = R'-9-BBN

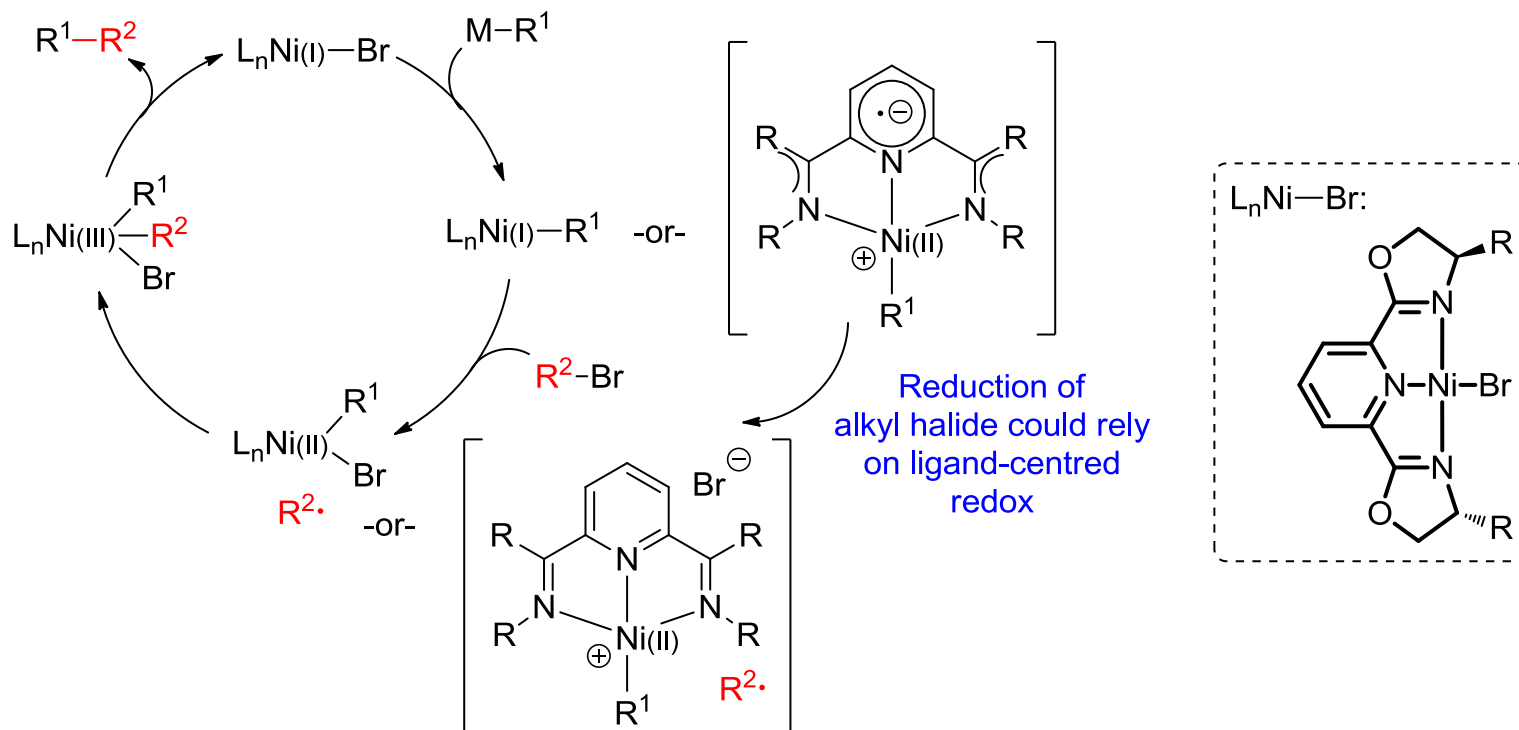
Efficient
Robust
General

up to 90% yield
up to 91% ee (5 examples)



Jiang, X.; Sakhivel, S.; Kulbitski, K.; Nisnevich, G.; Gandelman, M.
J. Am. Chem. Soc. **2014**, *136*, 9548

Cross-coupling of sp^3 Halides



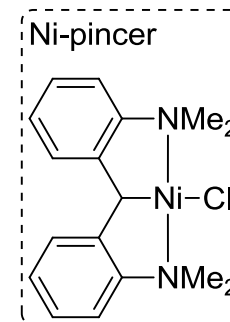
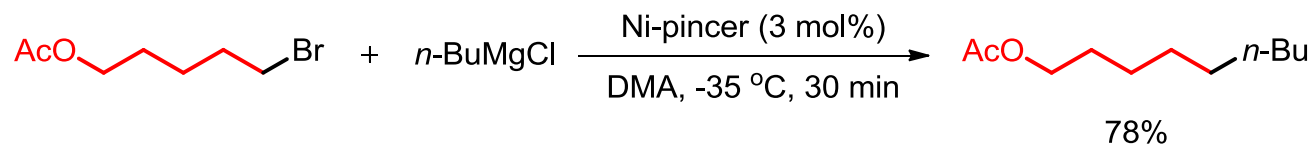
Proposed mechanisms for $C(sp^3)-C(sp^3)$ cross-coupling

Vicic, D. A. *J. Am. Chem. Soc.* **2006**, *128*, 13175.

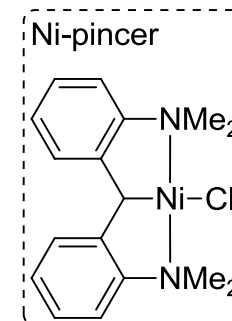
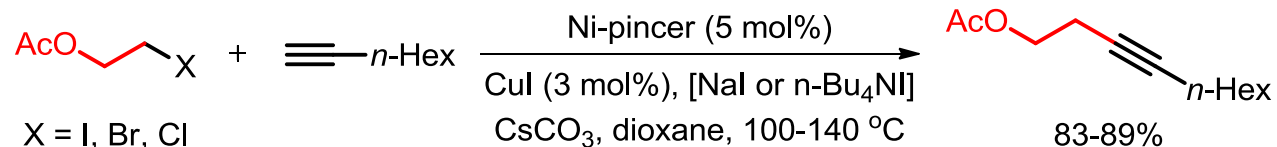
Dudnik, A. S.; Fu, G. C. *J. Am. Chem. Soc.* **2012**, *134*, 10693.

Zultanski, S. L.; Fu, G. C. *J. Am. Chem. Soc.* **2013**, *135*, 624.

Cross-coupling of sp^3 Halides

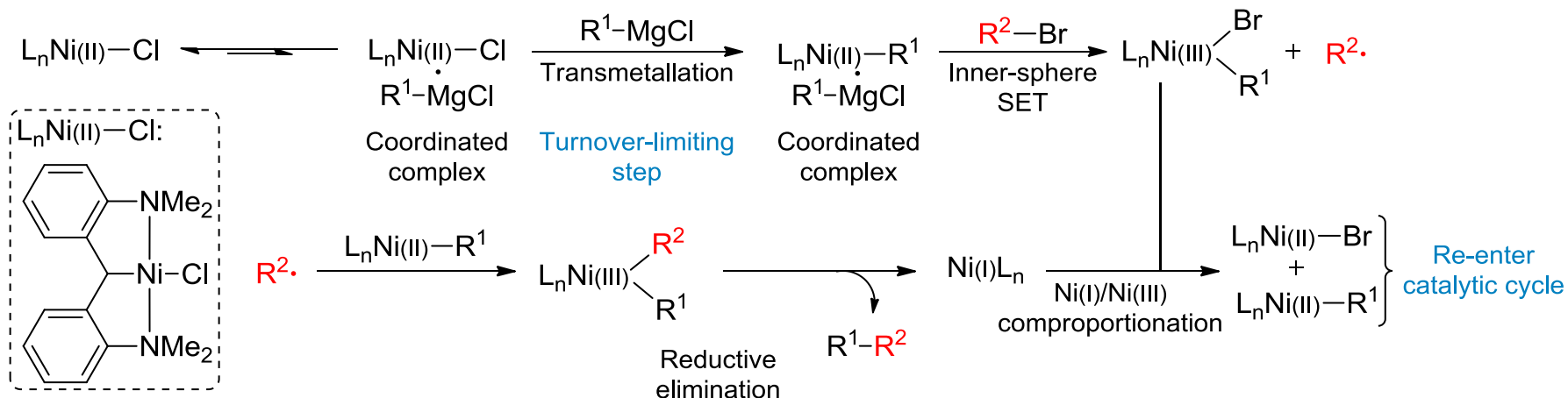


Csok, Z.; Vechorkin, O.; Scopelliti, R.; Hu, X. *J. Am. Chem. Soc.* **2008**, *130*, 8156.
Vechorkin, O.; Hu, X. *Angew. Chem. Int. Ed.* **2009**, *48*, 2937.



Vechorkin, O.; Barmaz, D.; Proust, V.; Hu, X. *J. Am. Chem. Soc.* **2009**, *131*, 12078.

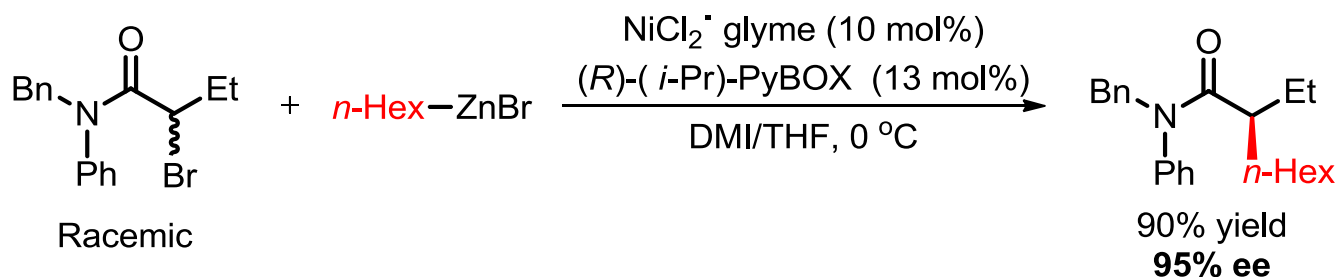
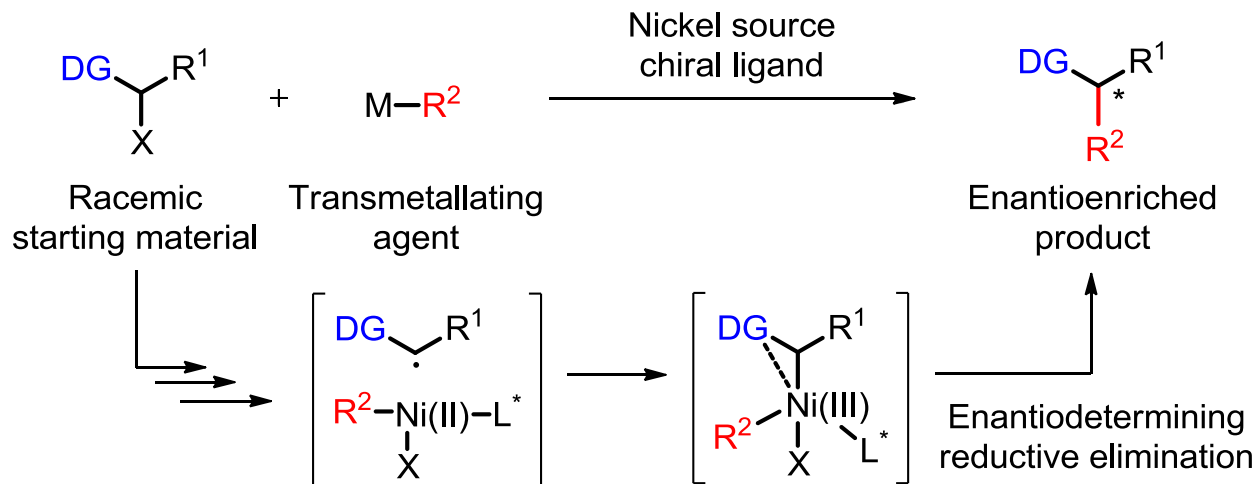
Cross-coupling of sp^3 Halides



Proposed mechanisms for $C(sp^3)-C(sp^3)$ cross-coupling

Breitenfeld, J.; Ruiz, J.; Wodrich, M. D.; Hu, X. *J. Am. Chem. Soc.* **2013**, *135*, 12004.

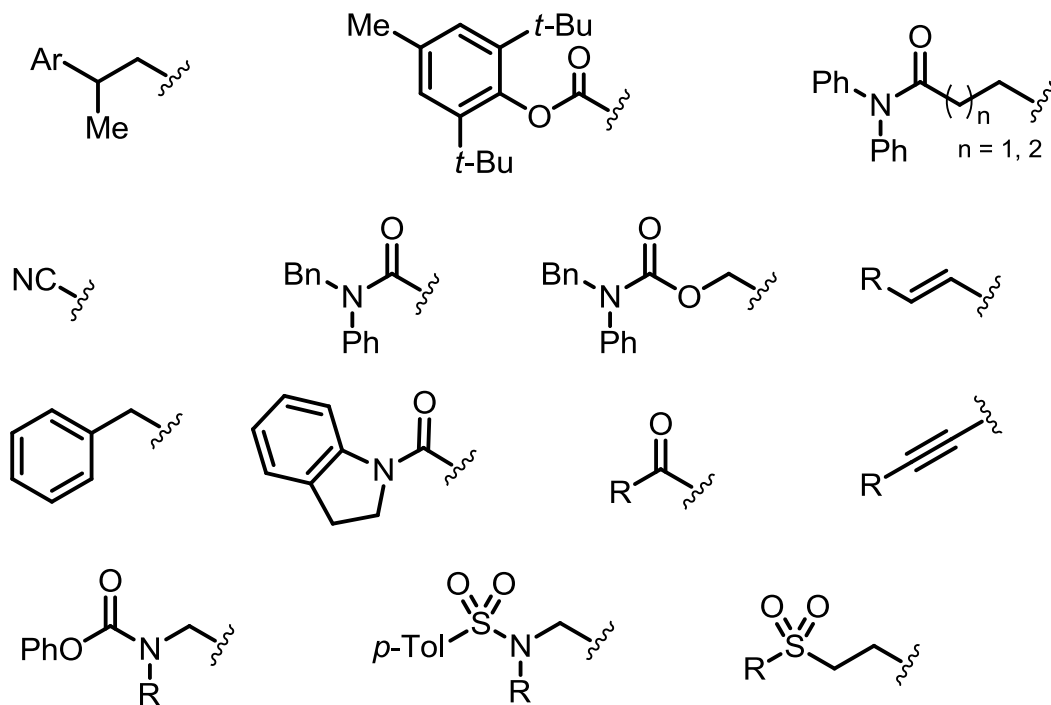
Cross-coupling of sp^3 Halides



Fischer, C.; Fu, G. C. *J. Am. Chem. Soc.* **2005**, *127*, 4594.

Cross-coupling of sp^3 Halides

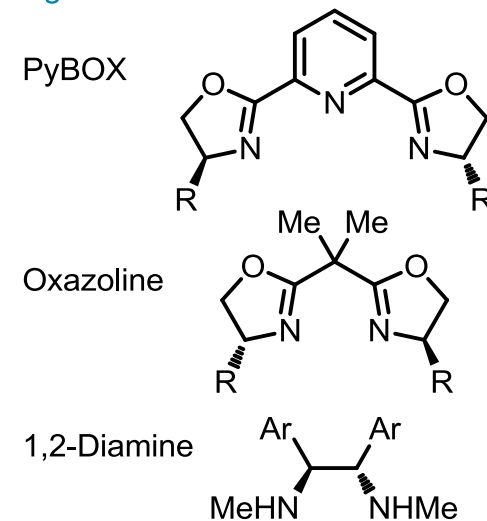
Directing groups



Transmetallating agents

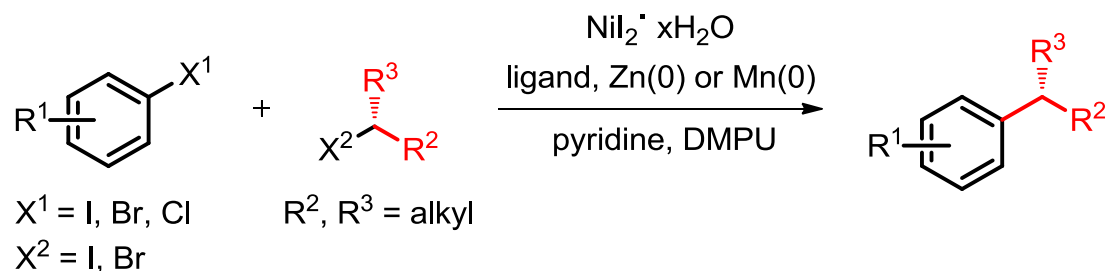
alkyl, aryl, alkenyl
zinc (Negishi), boron (Suzuki)
silicon (Hiyama), magnesium (Kumada),
and zirconium reagents

Ligand classes

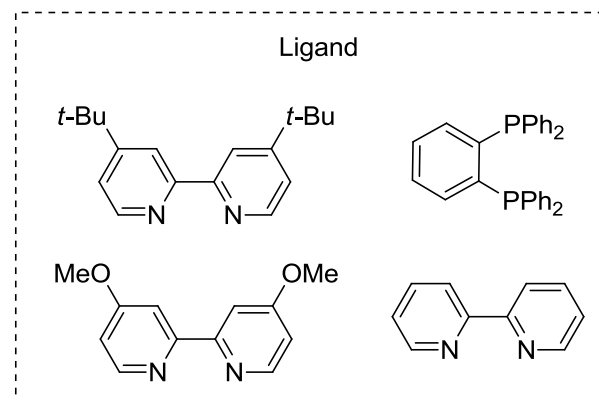
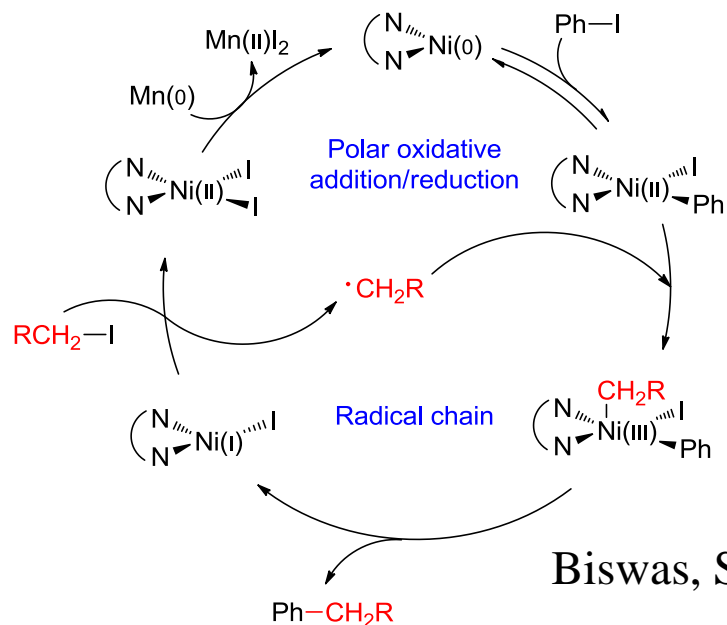


Reductive Cross-coupling

Challenge: the differentiation between the two electrophiles, suppress homodimerization of either component.

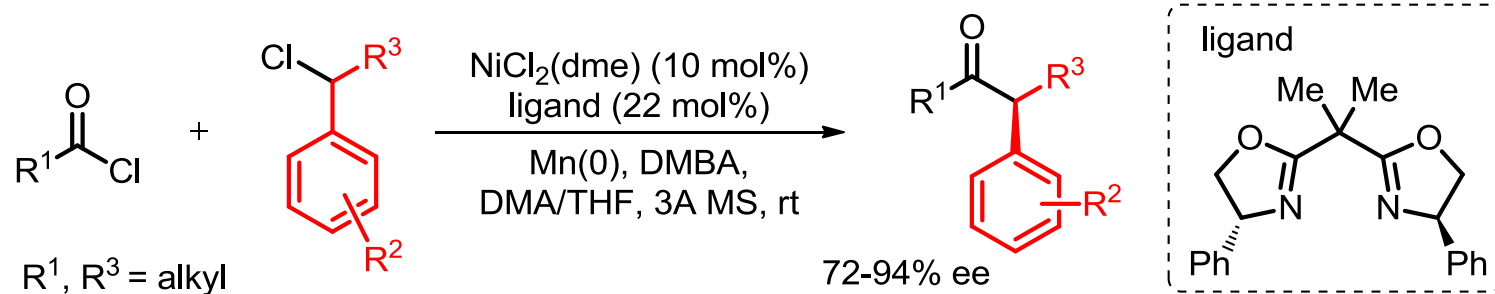


Everson, D. A.; Shrestha, R.; Weix, D. J. *J. Am. Chem. Soc.* **2010**, *132*, 920.



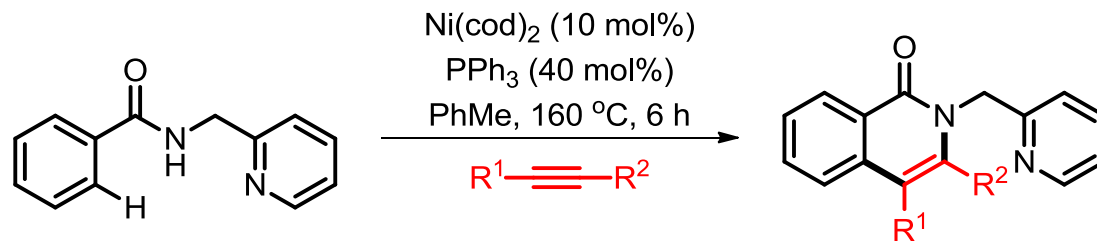
Biswas, S.; Weix, D. J. *J. Am. Chem. Soc.* **2013**, *135*, 16192.

Reductive Cross-coupling

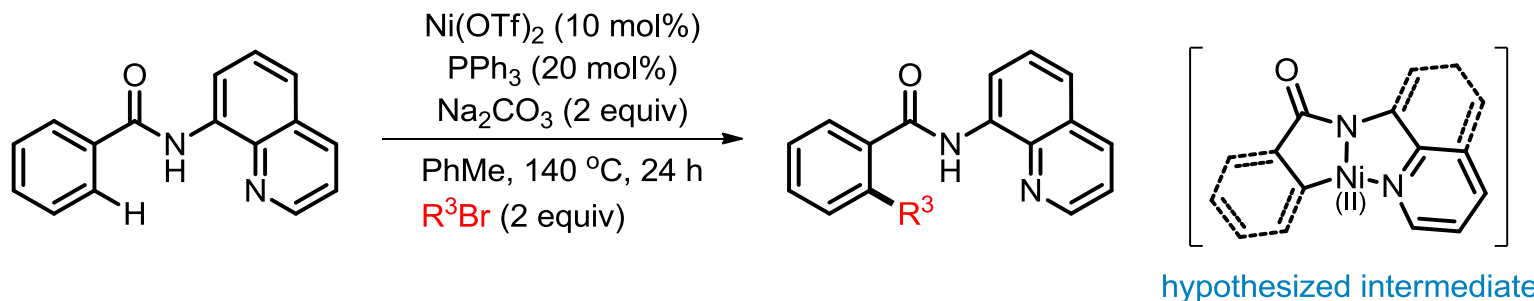


Cherney, A. H.; Kadunce, N. T.; Reisman, S. E. *J. Am. Chem. Soc.* **2013**, *135*, 7442.

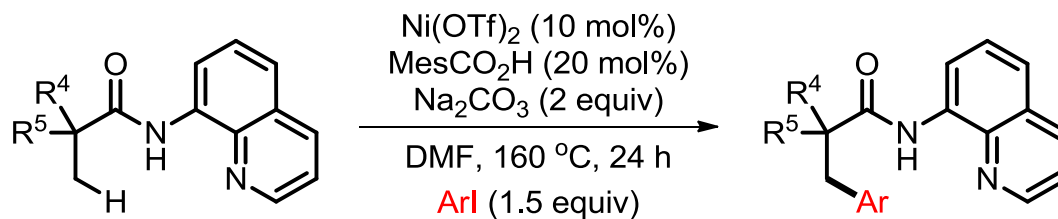
C–H Activation



Shiota, H.; Ano, Y.; Aihara, Y.; Fukumoto, Y.; Chatani, N. *J. Am. Chem. Soc.* **2011**, *133*, 14952.



Aihara, Y.; Chatani, N. *J. Am. Chem. Soc.* **2013**, *135*, 5308.

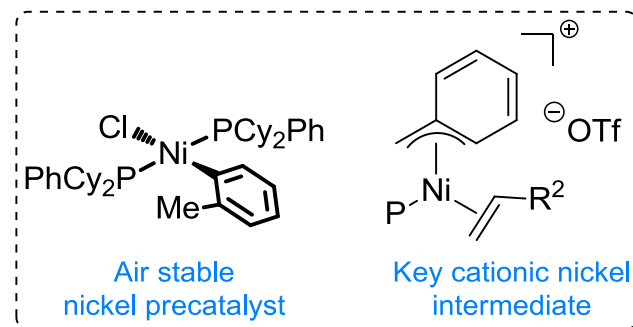
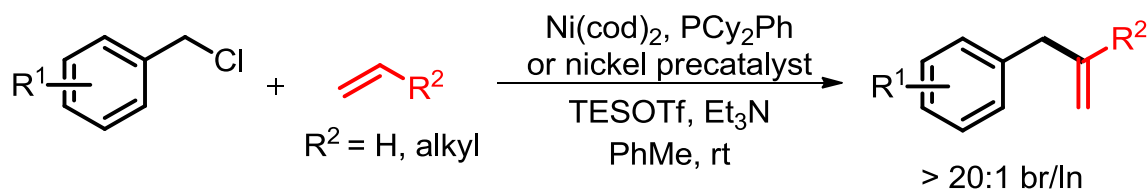


Aihara, Y.; Chatani, N. *J. Am. Chem. Soc.* **2013**, *136*, 898.

Heck Reaction

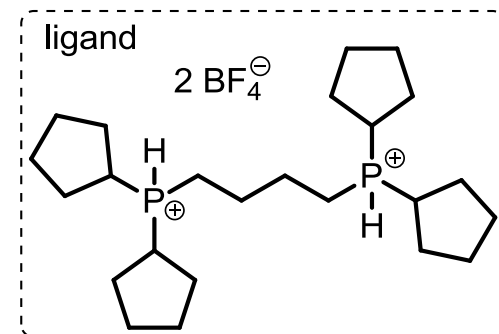
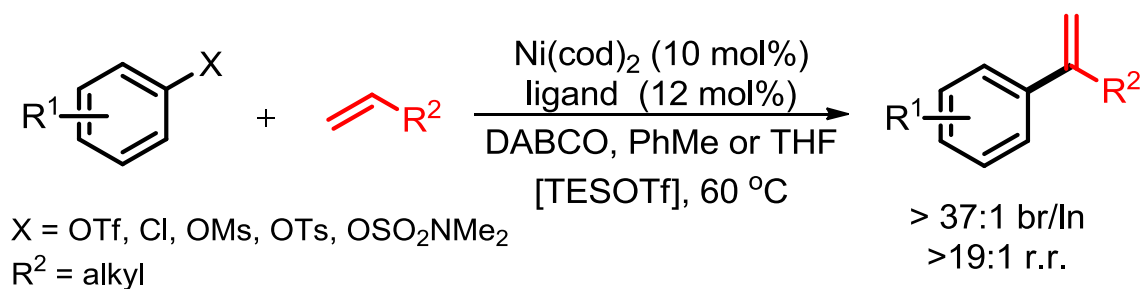


Gøgsig, T. M.; Kleimark, J.; Nilsson Lill, S. O.; Korsager, S.; Lindhardt, A. T.; Norrby, P.-O.; Skrydstrup, T. *J. Am. Chem. Soc.* **2011**, *134*, 443.



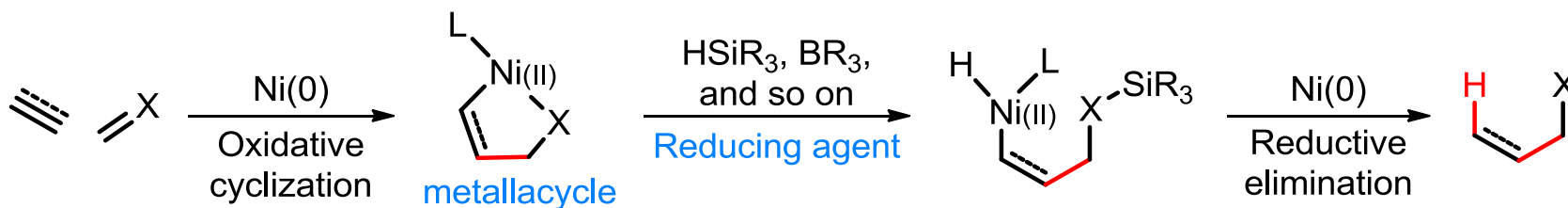
Matsubara, R.; Gutierrez, A. C.; Jamison, T. F. *J. Am. Chem. Soc.* **2011**, *133*, 19020.
Standley, E. A.; Jamison, T. F. *J. Am. Chem. Soc.* **2013**, *135*, 1585.

Heck Reaction



Tasker, S. Z.; Gutierrez, A. C.; Jamison, T. F. *Angew. Chem. Int. Ed.* **2014**, *53*, 1858.

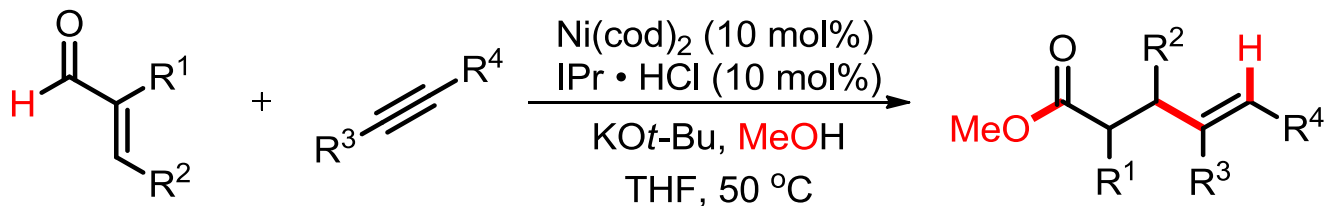
Reductive Coupling



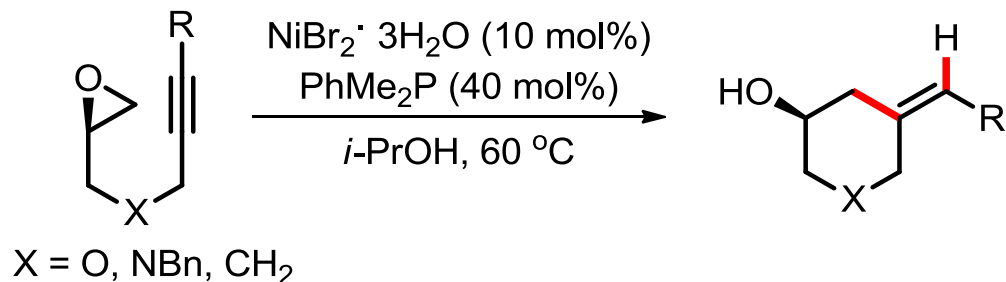
Moslin, R. M.; Miller-Moslin, K.; Jamison, T. F. *Chem. Commun.* **2007**, 4441.

Ng, S.-S.; Ho, C.-Y.; Jamison, T. F. *Pure Appl. Chem.* **2008**, 80, 929.

Reductive Coupling



Herath, A.; Li, W.; Montgomery, J. *J. Am. Chem. Soc.* **2007**, *130*, 469.



Beaver, M. G.; Jamison, T. F. *Org Lett.* **2011**, *13*, 4140.

Summary and Outlook

Outlook:

- Design new catalysts, develop new transformations. Such as C-F formation.
- Further in the area of C(sp³)–C(sp³) bond formation, particularly in expansion of substrate scope and application to the synthesis of complex molecules.
- in the development of low-cost, air-stable, and easier-to-handle sources of nickel for catalysis.
- Gain more recognition, not as an inexpensive substitute for palladium, but rather as possessing a number of inherent properties that provide a complement to catalysis by other metals

Thank you for your attention !