

Catalytic carbene insertion into C-H bonds

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Reaction Mechanism

Catalysts and reactions

Summary and outlook



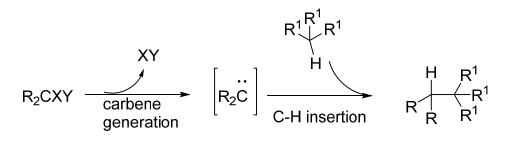
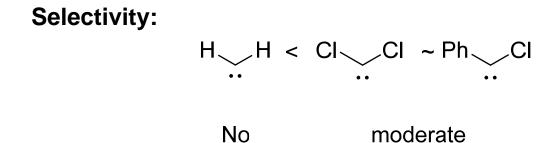


Figure1. Generalized carbene generation and carbon-hydrogen insertion



Thermally or photochemically generated carbenes show no potential for synthetic uses.

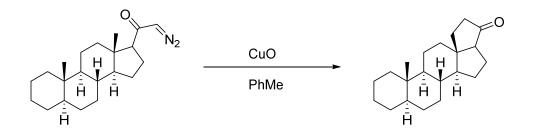


Figure 2. The first transition metal participate carbene insertion into C-H bond reaction

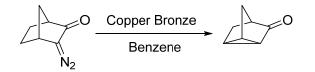
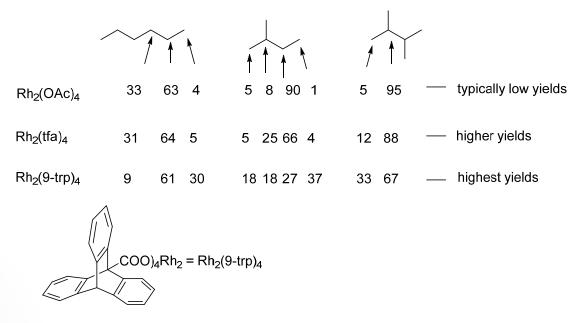


Figure 3. Copper catalysed C-H insertion in rigid system

The breakthrough that brought carbon-hydrogen insertion reactions into the realm of viable synthetic applicability was the report of the Teyssie group of intermolecular carbon-hydrogen insertion reactions of ethyl diazoacetate with alkanes, catalyzed by dirhodium(II) tetraacetate and rhodium carboxylate derivatives (Scheme 1)



Scheme 1. For reactions with ethyl diazoacetate, arrows difine site of C-H insertion

since the early 1990s, the majority of reports have concerned enantioselective C–H insertion reactions, Early work in this area was conducted by Taber and Wenkert, who demonstrated that C–H insertion reactions may occur with high levels of regio- and diastereoselectivities.

Some basic data:

1. The activation enthalpy of the Rh₂(OAc)₄-catalyzed nitrogen extrusion reaction of ethyl diazoacetate is the rate-limiting step of the catalytic cycle for secondary C-H insertion

2. Reactivity: tertiary > secondary >> primary, and the enhanced reactivity Of a C-H bond adjacent to a heteroatom.

3. Insertion into a C-H bond occurred with retention of configuration

Reaction Mechanism

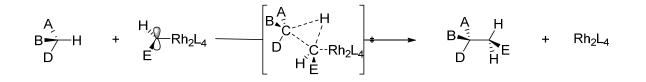


Figure 4. Transition state models of C-H bond activation with dirhodiumcarbene Complex proposed by Doyle



Reaction Mechanism

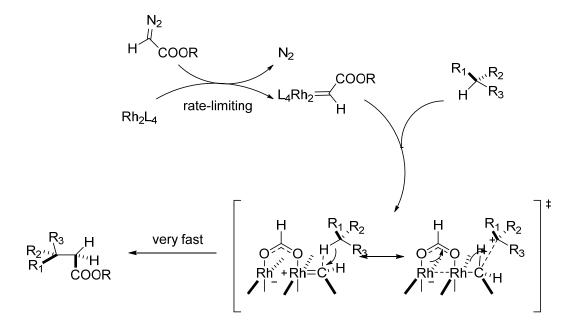
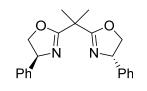
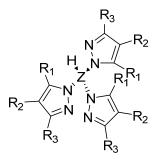


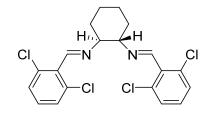
Figure 5. Mechanism fo C-H bond insertion into metal carbene

Copper catalyst and related reactions

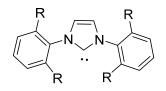


Bis(oxazoline)





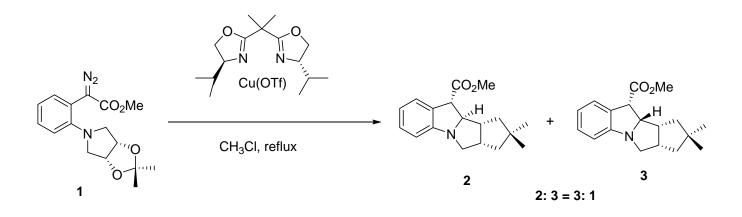
Schiff base



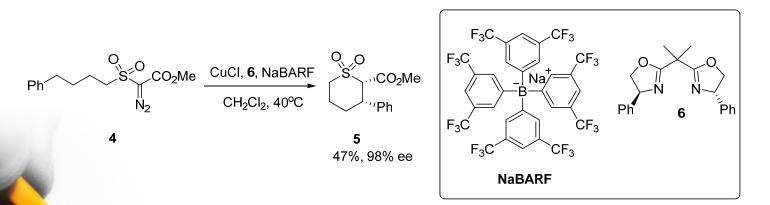
NHC

Tp^x (Z=B) Tpm^x (Z=CH)

Scheme 2. Ligands used in copper(I) catalyst

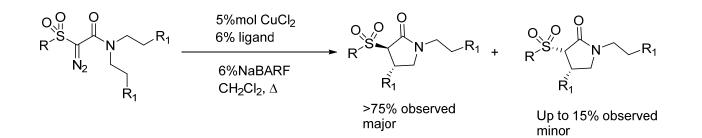


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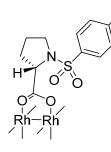


Maguire, A. R. Org. Biomol. Chem. 2014, 12, 7612-7628.

High yields (70–90%) and moderate to high enantioselectivities (up to 82% ee) were observed

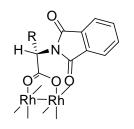
Rhodium catalysts and related reactions

Chiral rhodium carboxylate catalysts



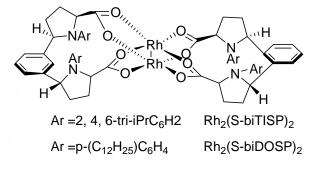
R=H, Rh₂(S-BSP)₄ R=C₁₂H₂₅, Rh₂(S-DOSP)₄

Proline complexes



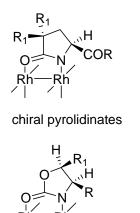
 $\begin{array}{l} \mathsf{R} = \mathsf{Me}, \ \mathsf{Rh}_2(\mathsf{S} \text{-} \mathsf{PTA})_4 \\ \mathsf{R} = \mathsf{C}(\mathsf{Et})3, \ \mathsf{Rh}_2(\mathsf{S} \text{-} \mathsf{PTTEA})_4 \end{array}$

Phthalimide derivatives

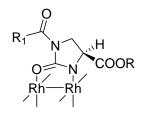


bridged prolinate comlexes

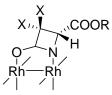
Chiral rhodium carboxamidate catalysts



chiral oxazolidinates



chiral imidazolidinates

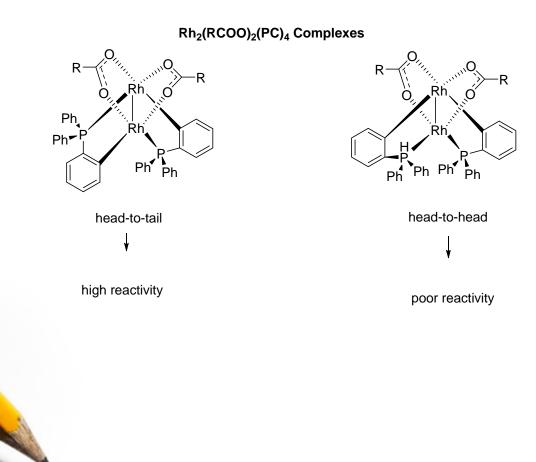


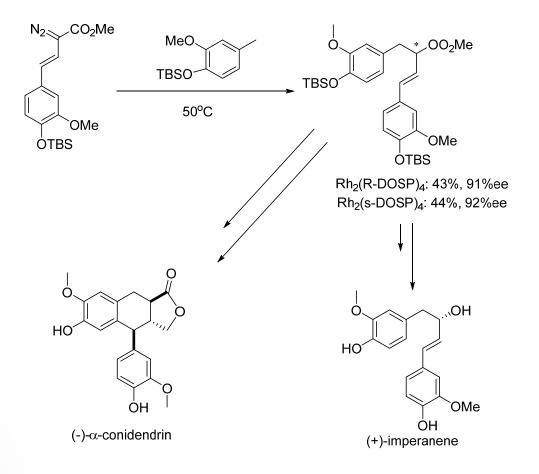
chiral azetidinates

These catalysts have since been much exploited and remain today the primary catalysts for enantioselective C-H insertion reactions of electron-withdrawing group substituted carbenoids derived from diazoacetamides and diazaoesters.

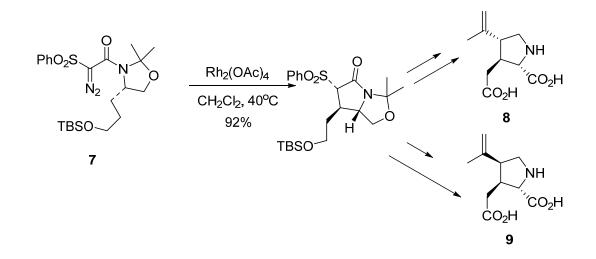
Hu Group

Chiral ortho-metalated rhodium complexes

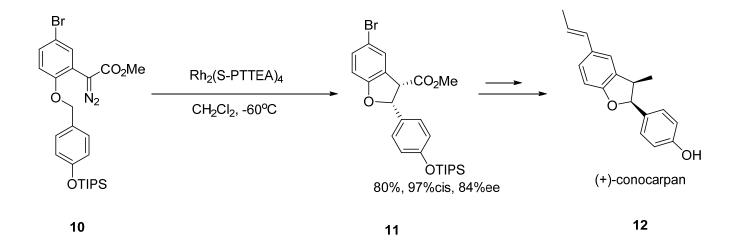




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Hu Group

- 1. A rough knowledge about carbene insertion into C-H bonds have been introduced
- 2. Dirhodium carboxylate and carboxamidate catalysts are most effective in these transformations.
- 3. While rhodium complexes remain the dominant catalysts for application in enantioselective C-H insertion reactions, developing alternative cheaper metal catalysts is a possible way for future studies.
- 4. Development of a catalyst system with general applicability across the spectrum of intramolecular and intermolecular C-H insertion reactions remains challenging.

Referrences

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