

# Divergent S- and C-Difluoromethylation of 2-Substituted Benzothiazoles

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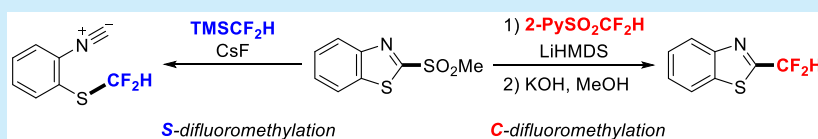
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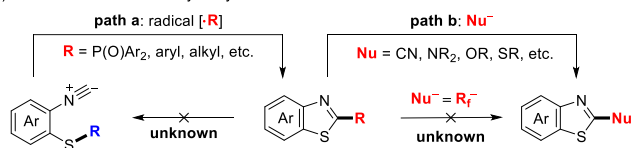


**ABSTRACT:** Two unprecedented and complementary synthetic strategies for S- and C-difluoromethylation of 2-substituted benzothiazoles have been developed by taking advantage of the remarkably different reactivity of  $\text{CF}_2\text{H}^-$  and  $2\text{-PySO}_2\text{CF}_2^-$  nucleophiles. A variety of structurally diverse difluoromethyl 2-isocyanoaryl sulfides and 2-difluoromethylated benzothiazoles were synthesized with these two new synthetic protocols.

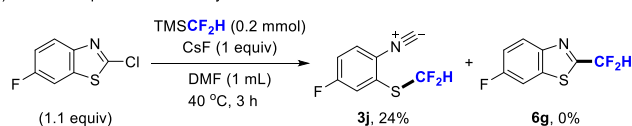
Isocyanides are versatile synthetic building blocks for the synthesis of nitrogen-containing molecules.<sup>1</sup> Their wide applications in multicomponent reactions,<sup>2</sup> transition-metal-catalyzed isocyanide insertions,<sup>3</sup> and radical cascade reactions<sup>4,5</sup> provide efficient access to an array of pharmaceutically relevant molecules. In particular, 2-isocyanoaryl thioethers could be used as radical acceptors in radical cyclization reactions to construct 2-substituted benzothiazoles (Scheme 1a, path a),<sup>6</sup> which are prevalently existing in various naturally occurring compounds and drugs.<sup>7</sup> However, despite the remarkable progress in the cyclization of 2-isocyanoaryl thioethers to 2-substituted benzothiazoles,<sup>6</sup> its reverse process, namely, the ring-opening reaction of 2-substituted benzothiazoles to give 2-isocyanoaryl thioethers, still remains unknown.

## Scheme 1. Transformations of 2-Isocyanoaryl Thioethers and 2-Substituted Benzothiazoles

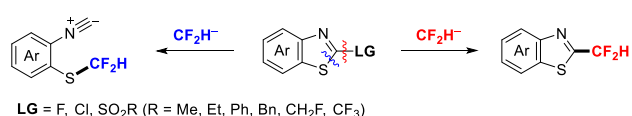
a) Reaction modes of 2-isocyanoaryl thioethers and 2-substituted benzothiazoles



b) Initial attempt on difluoromethylation of 2-substituted benzothiazoles



c) **This work:** selective S- and C-difluoromethylation of 2-substituted benzothiazoles



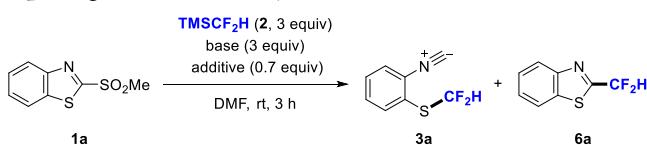
To date, the known reaction mode of 2-substituted benzothiazoles is the nucleophilic aromatic substitution ( $\text{S}_{\text{N}}\text{Ar}$ ) at the C-2 position with a series of C-, N-, O- and S-nucleophiles (Scheme 1a, path b).<sup>8</sup> However, to the best of our knowledge, the  $\text{S}_{\text{N}}\text{Ar}$  reaction between a fluoroalkyl nucleophile and a 2-substituted benzothiazole has never been reported. Difluoromethyl group ( $\text{CF}_2\text{H}$ ) can serve as a lipophilic hydrogen bond donor, and a bioisostere of OH or SH functionality.<sup>9</sup> In this context, we initially attempted the  $\text{S}_{\text{N}}\text{Ar}$ -type nucleophilic difluoromethylation of 2-substituted benzothiazoles with (difluoromethyl)trimethylsilane<sup>10</sup> ( $\text{TMSCF}_2\text{H}$ , **2**). To our surprise, the expected product 2-difluoromethyl benzothiazole (**6g**) was not observed, and instead, difluoromethyl 2-isocyanoaryl sulfide (**3j**) was obtained in 24% yield (Scheme 1b). It is likely that an unprecedented S-difluoromethylation-ring-opening elimination tandem occurred.<sup>11</sup> This unexpected result intrigued us to investigate the nucleophilic difluoromethylation of 2-substituted benzothiazoles, especially the selectivity between  $\text{S}_{\text{N}}\text{Ar}$  and aromatic ring opening reactions (Scheme 1c).

At the onset of our investigation, we chose 2-methanesulfonyl benzothiazole **1a** as a model substrate and  $\text{TMSCF}_2\text{H}$  (**2**) as the difluoromethyl anion precursor, and the reaction conditions were screened (Table 1). When a mixture of **1a** (0.2 mmol), **2** (3 equiv), CsF (3 equiv) in DMF were stirred at room temperature, neither ring-opening product **3a** nor  $\text{S}_{\text{N}}\text{Ar}$  product **6a** was observed (entry 1). It is interesting that when

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**Table 1. Optimization of the Reaction Conditions of Ring-Opening S-Difluoromethylation of Benzothiazole 1a**

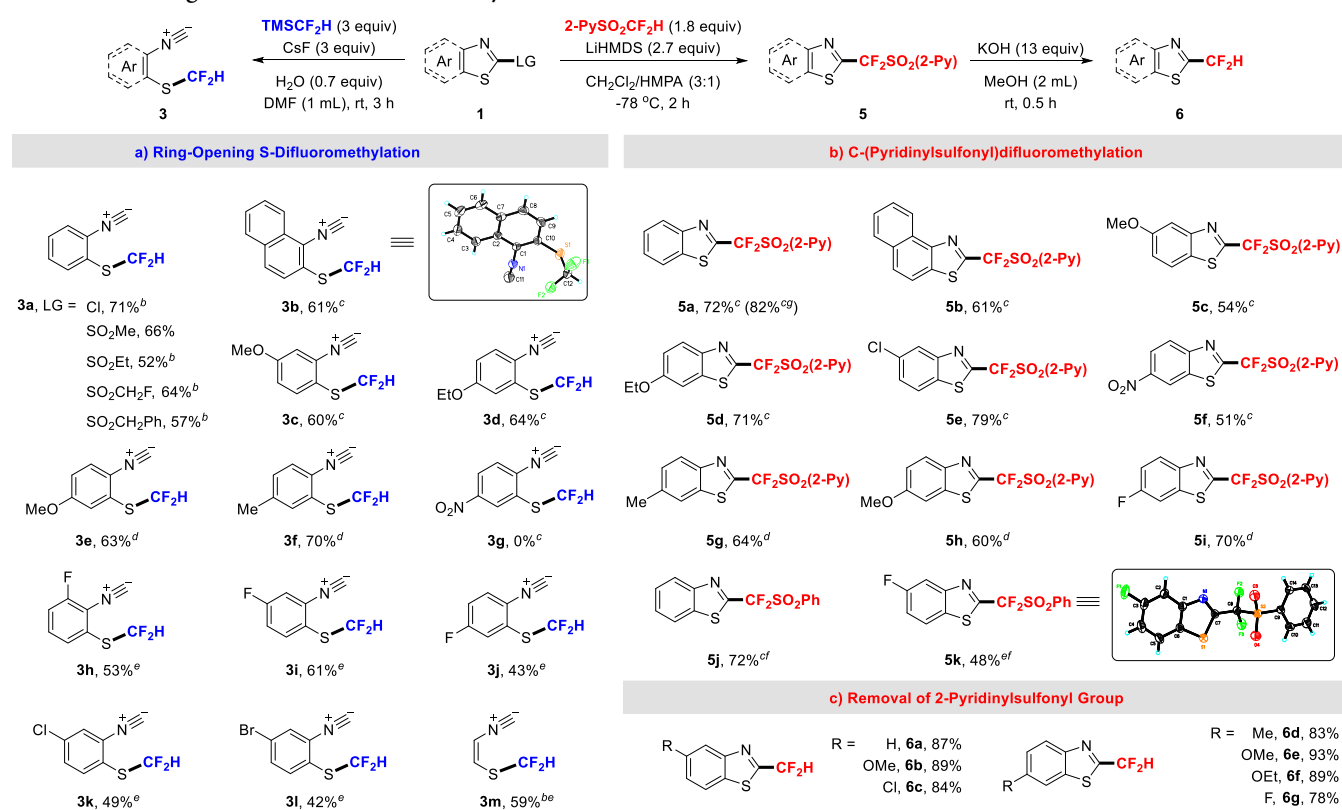
entry <sup>a</sup>	base	additive	yield (%) <sup>b</sup>	
			3a	6a
1	CsF	none	0	0
2	CsF	H <sub>2</sub> O	70 (66)	0
3	CsF	MeOH	53	0
4	TBAT	H <sub>2</sub> O	33	0
5	KF	H <sub>2</sub> O	6	0
6	K <sub>2</sub> CO <sub>3</sub>	H <sub>2</sub> O	14	0
7	CsOH	H <sub>2</sub> O	9	0

<sup>a</sup>Reaction conditions: **1a** (0.2 mmol, 1 equiv), **2** (0.6 mmol, 3 equiv), base (3 equiv), additive (0.7 equiv), DMF (1 mL), rt, 3 h. <sup>b</sup>Yields were determined by <sup>19</sup>F NMR spectroscopy using PhCF<sub>3</sub> as an internal standard, and the isolated yield was shown in parentheses.

0.7 equiv of water was added, 70% yield of the ring-opening product **3a** was formed (entry 2). To figure out the role of water, we replaced water with another proton source methanol, and found that **3a** was formed in slightly lower yield (53%; entry 3). Furthermore, when a mixture of **3a** (0.2 mmol), TMSCF<sub>2</sub>H (3 equiv), CsF (3 equiv) and DMF (1 mL) was stirred at room temperature for 3 h in the absence of water,

compound **3a** completely decomposed (Scheme S1 in Supporting Information). These results indicate that water serves as a proton source and quenches the unreacted difluoromethyl anion (forming CH<sub>2</sub>F<sub>2</sub>), which diminishes the decomposition of **3a**. The amount of water was carefully screened, and 0.7 equiv of water was identified as optimal for the reaction (Table S2 in Supporting Information). In addition to CsF, other Lewis base activators such as TBAT, KF, K<sub>2</sub>CO<sub>3</sub>, and CsOH were tested, but all showed lower efficiency than CsF in activating the Si–CF<sub>2</sub>H bond cleavage in DMF at room temperature (entries 4–7). Notably, in all cases, no formation of S<sub>N</sub>Ar-type C-difluoromethylation product **6a** was observed during the current ring-opening S-difluoromethylation reaction (Table 1).

With the standard reaction conditions in hand (Table 1, entry 2), the substrate scope of the ring-opening S-difluoromethylation reaction with TMSCF<sub>2</sub>H was investigated (Scheme 2a). First, the leaving group ability of different groups at C-2 position of **1** was examined (LG = Cl, SO<sub>2</sub>Me, SO<sub>2</sub>Et, SO<sub>2</sub>CH<sub>2</sub>F, and SO<sub>2</sub>CH<sub>2</sub>Ph), and in all cases the reaction proceeded smoothly, affording product **3a** in moderate to good yields. Second, we chose 2-(methanesulfonyl)naphthothiazole as a substrate in the reaction, and product **3b** was obtained in 61% yield. The structure of **3b** was confirmed by its single crystal X-ray analysis. Third, benzothiazoles bearing electron-donating groups (such as methyl, methoxyl and ethoxyl) are amenable to the present ring-opening S-difluoromethylation reaction, with the corresponding products **3c–3f** being formed in 60–70% yields. However, no desired product **3g** was

**Scheme 2. Divergent S- and C-Difluoromethylation of 2-Substituted Benzothiazoles<sup>a</sup>**

<sup>a</sup>Isolated yields. <sup>b</sup>Yields were determined by <sup>19</sup>F NMR spectroscopy using trifluorotoluene as an internal standard. <sup>c</sup>LG = SO<sub>2</sub>Me. <sup>d</sup>LG = SO<sub>2</sub>Ph. <sup>e</sup>LG = Cl; 2.0 equiv of H<sub>2</sub>O were used in S-difluoromethylation. <sup>f</sup>PhSO<sub>2</sub>CF<sub>2</sub>H was used instead of 2-PySO<sub>2</sub>CF<sub>2</sub>H, THF (1.2 mL) was used instead of CH<sub>2</sub>Cl<sub>2</sub>. <sup>g</sup>6 mmol scale reaction.

detected when a nitrated benzothiazole was subjected to the standard reaction conditions. Fourth, in cases of 2-chlorobenzothiazoles, the halogen substituents (such as F, Cl and Br) on the aromatic ring were compatible with the reaction conditions (see **3h–3l**). Finally, 2-chlorothiazole was also able to participate in the reaction, and target product **3m** was formed in 59% yield (determined by NMR).

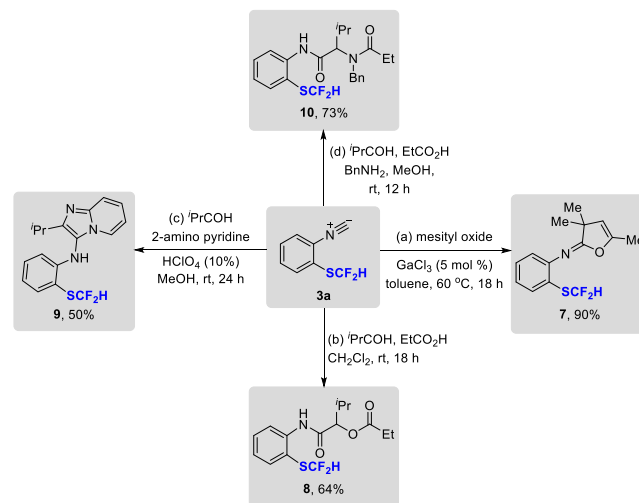
Encouraged by the aforementioned ring-opening S-difluoromethylation of 2-substituted benzothiazoles with  $\text{TMSCF}_2\text{H}$ , we further applied other fluorinated nucleophiles in the reaction with 2-substituted benzothiazoles. Difluoromethyl 2-pyridyl sulfone (2-PySO<sub>2</sub>CF<sub>2</sub>H), a reagent developed by us, is now commercially available and widely used for fluoroalkylation and fluoroolefination reactions.<sup>12</sup> We were surprised to find, when 2-(methanesulfonyl)benzothiazole **1a** reacted with 2-PySO<sub>2</sub>CF<sub>2</sub><sup>−</sup> (derived from 2-PySO<sub>2</sub>CF<sub>2</sub>H and LiHMDS) in CH<sub>2</sub>Cl<sub>2</sub>/HMPA at −78 °C for 2 h, no ring-opening S-(2-pyridinesulfonyl)difluoromethylation product was formed, but S<sub>N</sub>Ar-type C-(2-pyridinesulfonyl)difluoromethylation product **5a** was obtained in 72% isolated yield (see Table S3 in Supporting Information and Scheme 2b). This result suggests that different  $\alpha$ -fluoro carbanions show remarkably different reactivity (chemoselectivity) in their reactions with 2-substituted benzothiazoles.

Next, we examined the generality of the current S<sub>N</sub>Ar-type C-(2-pyridinesulfonyl)difluoromethylation of **1** (Scheme 2b). First, the reaction was not sensitive to leaving groups at C-2 position of benzothiazoles **1**. When chloro, methanesulfonyl, and benzenesulfonyl groups were used as the leaving groups, the S<sub>N</sub>Ar-type C-fluoroalkylation reactions proceeded smoothly to yield the corresponding products in 48–79% yields (**5a–5k**). The current C-fluoroalkylation reaction was applied in gram-scale synthesis, and product **5a** was obtained in 82% isolated yield (1.6 g). Second, the reaction was able to tolerate different substituents on **1**, including methyl, methoxyl, ethoxyl, chloro, and nitro groups (**5c–5i**). Third, when we replaced 2-PySO<sub>2</sub>CF<sub>2</sub>H with PhSO<sub>2</sub>CF<sub>2</sub>H, the S<sub>N</sub>Ar-type C-benzenesulfonyl difluoromethylation products **5j** and **5k** were also successfully formed in 72% and 48% yields, respectively. With products **5** in hand, we performed the base-promoted selective desulfonylation to transform **5** to difluoromethylated compounds **6** (Scheme 2c, also see details in Table S4 in Supporting Information). In the presence of KOH (13 equiv) in methanol at room temperature, **5** were readily converted into **6** within 30 min in 78–93% yields (Scheme 2c). Given the results shown in Scheme 2, it is intriguing that the remarkably different reactivity of CF<sub>2</sub>H<sup>−</sup> and 2-PySO<sub>2</sub>CF<sub>2</sub><sup>−</sup> have enabled two complementary synthetic strategies for highly chemoselective S- and C-difluoromethylation of 2-substituted benzothiazoles **1**.

To demonstrate the synthetic utility of the current synthetic protocol, we applied **3a** as a valuable building block in multicomponent reactions. First, **3a** was used in a GaCl<sub>3</sub>-catalyzed [4 + 1] cycloaddition with mesityl oxide, which afforded SCF<sub>2</sub>H-substituted unsaturated lactone derivative **7** in 90% yield (Scheme 3a).<sup>13</sup> Furthermore, **3a** was also applied in Passerini, Gröbcke–Blackburn–Bienaymé, and Ugi reactions, giving products **8–10** in moderate to good yields (Scheme 3b–d).

In summary, we have successfully developed two unprecedented and complementary synthetic strategies for divergent S- and C-difluoromethylation of 2-substituted benzothiazoles by taking advantage of the remarkably different reactivity of

Scheme 3. Synthetic Applications of **3a**



CF<sub>2</sub>H<sup>−</sup> and 2-PySO<sub>2</sub>CF<sub>2</sub><sup>−</sup> nucleophiles. A variety of structurally diverse difluoromethyl 2-isocyanophenyl sulfides **3** and 2-difluoromethylated benzothiazoles **6** were synthesized with these two new synthetic protocols. Our study uncovers the unique reactivity of CF<sub>2</sub>H<sup>−</sup>, which promises to stimulate further development of new difluoromethylation reactions.

## ■ ASSOCIATED CONTENT

### Supporting Information

The Supporting Information is available free of charge at <https://pubs.acs.org/doi/10.1021/acs.orglett.1c03267>.

Full characterization, copies of all spectral data, crystallographic data of CCDC 2031759 (**3b**) and CCDC 2031760 (**5k**), and experimental procedures (PDF)

### Accession Codes

CCDC 2031759–2031760 contain the supplementary crystallographic data for this paper. These data can be obtained free of charge via [www.ccdc.cam.ac.uk/data\\_request/cif](http://www.ccdc.cam.ac.uk/data_request/cif), or by emailing [data\\_request@ccdc.cam.ac.uk](mailto:data_request@ccdc.cam.ac.uk), or by contacting The Cambridge Crystallographic Data Centre, 12 Union Road, Cambridge CB2 1EZ, UK; fax: +44 1223 336033.

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### Author Contributions

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### Notes

The authors declare no competing financial interest.

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